

## Chapter 10 Nuclear Chemistry

**Section 10.2 Rates of Nuclear Decay****(pages 298–301)**

*This section discusses half-lives and explains how nuclear decay can be used to estimate the age of objects.*

**Reading Strategy (page 298)**

**Identifying Details** As you read, complete the concept map below to identify details about radiocarbon dating. For more information on this Reading Strategy, see the **Reading and Study Skills** in the **Skills and Reference Handbook** at the end of your textbook.

**Half-life (pages 299–300)**

1. A nuclear decay rate describes how fast nuclear changes take place in a radioactive substance.
2. Is the following sentence true or false? All radioisotopes decay at the same rate. false
3. Describe a half-life. A half-life is the time required for one half of a sample of a radioisotope to decay.
4. Circle the letter that describes a sample of a radioisotope after two half-lives.
  - a. One eighth of the original sample is unchanged.
  - b.** One quarter of the original sample is unchanged.
  - c. Half of the original sample is unchanged.
  - d. Three quarters of the original sample is unchanged.
5. Circle the letter of the correct answer. Iodine-131 has a half-life of 8.07 days. What fraction of a sample of iodine-131 is left unchanged after 16.14 days?
 

a. $\frac{1}{2}$	<b>b. <math>\frac{1}{4}</math></b>
c. $\frac{1}{8}$	d. $\frac{1}{16}$
6. Is the following sentence true or false? Like chemical reaction rates, nuclear decay rates vary with the conditions of reaction.  
false

## Section 3 Rates Of Nuclear Decay Answers

**Michael J. Findley, Mary C. Findley**



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**Instructor's Guide IG 11.21. Radiological Monitoring Instructor's Guide** United States. Civil Defense Office, 1963

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**Radiological Monitoring** United States. Office of Civil Defense, 1963

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