Factors Affecting the Rate of a Chemical Reaction

The following events must occur before a reaction can proceed:

- 1. The reactant particles must collide with each other.
- The collisions must be of enough energy to overcome the 'energy barrier', called the activation energy (more about this on the next slide).
- The reactants must form new bonds to produce products.

This presentation is useful as a refresher for those of you starting Unit 2 of the A-level chemistry course.

Section 4 Reinforcement Chemical Reaction Rates And Energy

United States. Department of Commerce. Office of Technical Services

Section 4 Reinforcement Chemical Reaction Rates And Energy:

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