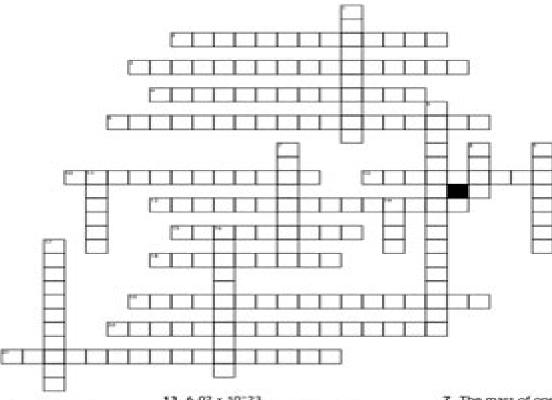
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## Stoichiometry/Chemical Reactions



2. The area of chemistry involving the calculations of quantities of substances involved in chemical reactions

3. A representation of a chemical reaction using symbols and numbers to show the relationships between the reactants and products

4. AB ---- A + B

6. The percent by mass of each element in a compound

10. The ratio of the actual yield as compared to the theoretical yield expressed as a percentage

12. A conversion factor derived from the coefficients of the quantities of substances involved in chemical reactions.

#### 13.6.02 x 10°23

15. The chemicals which are put into a chemical reaction and are on the left side of the reaction

18. A + 8 ----- AB

19. AB + CD ----- AD + CB

20. The substance that runs out in a chemical reaction, thus controlling the amount of product(s):

21. The amount of product that could form based on a balanced chemical equation

Charwon 1. A compound containing carbon and hydrogen is burned in the presence of

5. A + BC ---- AC + B

7. The mass of one mole of a substance

8. The measurement of the mass of the

substances in a chemical reaction

9. The chemicals present on the right side of a chemical reaction, that are only present after the the chemical reaction has begun

11. The reactant that is leftover after a reaction comes to completion

14. The amount of a substance

16. The amount of product that is actually formed when a reaction is carried out in the

17. The number in front of a balanced formula showing how much of that reactant. or product is present.

#### Word Bank

moteratio. percent composition percent yield. actual yield reactants

synthesis double replacement chemical equation limiting reactant

stoichiometry. avogadros number theoretical yield excess.

molarmass. coefficient combustion giraim.

decomposition single replacement mote products

# **Section Chemical Formulas And Equations Crossword Answers**

**American Chemical Society** 

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