## Section 9.3

## Limiting Reactants and Percent Yield

## Method 2:

## Calculations Involving a Limiting Reactant

## Steps for Solving Stoichiometry Problems Involving Limiting Reactants

- Step 1 Write and balance the equation for the reaction.
- Step 2 Convert known masses of reactants to moles.
- Step 3 Using the numbers of moles of reactants and the appropriate mole ratios, determine which reactant is limiting.
- Step 4 Using the amount of the limiting reactant and the appropriate mole ratios, compute the number of moles of the desired product.
- Step 5 Convert from moles of product to grams of product, using the molar mass (if this is required by the problem).

# **Section Review 9 3 Limiting Reactants And Percent Yield**

**Morris Hein, Susan Arena** 

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