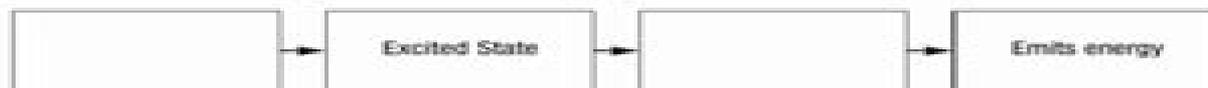


Chapter 4 Atomic Structure

Section 4.3 Modern Atomic Theory**(pages 113–118)***This section focuses on the arrangement and behavior of electrons in atoms.***Reading Strategy (page 113)**

Sequencing After you read, complete the description in the flow chart below of how the gain or loss of energy affects electrons in atoms. For more information on this Reading Strategy, see the **Reading and Study Skills** in the **Skills and Reference Handbook** at the end of your textbook.

**Bohr's Model of the Atom (pages 113–116)**

- Circle the letter of the sentence that tells how Bohr's model of the atom differed from Rutherford's model.
 - Bohr's model focused on the nucleus.
 - Bohr's model focused on the protons.
 - Bohr's model focused on the neutrons.
 - Bohr's model focused on the electrons.
- Is the following sentence true or false? In Bohr's model of the atom, electrons have a constant speed and move in fixed orbits around the nucleus. _____
- What can happen to an electron in an atom when the atom gains or loses energy? _____

- What evidence do scientists have that electrons can move from one energy level to another? _____

- Is the following sentence true or false? When electrons release energy, some of the energy may be released as visible light.

Electron Cloud Model (page 116)

- Is the following sentence true or false? Bohr's model was correct in assigning energy levels to electrons. _____
- When trying to predict the locations and motions of electrons in atoms, scientists must work with _____.
- What is an electron cloud? _____

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**Anatole Gregory Mazour, John M.
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Table of Contents Section Review Development Of The Atomic Theory Answers

1. Understanding the eBook Section Review Development Of The Atomic Theory Answers
 - The Rise of Digital Reading Section Review Development Of The Atomic Theory Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Section Review Development Of The Atomic Theory Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Section Review Development Of The Atomic Theory Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Section Review Development Of The Atomic Theory Answers
 - Personalized Recommendations
 - Section Review Development Of The Atomic Theory Answers User Reviews and Ratings
 - Section Review Development Of The Atomic Theory Answers and Bestseller Lists
5. Accessing Section Review Development Of The Atomic Theory Answers Free and Paid eBooks
 - Section Review Development Of The Atomic Theory Answers Public Domain eBooks
 - Section Review Development Of The Atomic Theory Answers eBook Subscription Services
 - Section Review Development Of The Atomic Theory Answers Budget-Friendly Options
6. Navigating Section Review Development Of The Atomic Theory Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Section Review Development Of The Atomic Theory Answers Compatibility with Devices
 - Section Review Development Of The Atomic Theory Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Section Review Development Of The Atomic Theory Answers
 - Highlighting and Note-Taking Section Review Development Of The Atomic Theory Answers
 - Interactive Elements Section Review Development Of The Atomic Theory Answers
8. Staying Engaged with Section Review Development Of The Atomic Theory Answers

- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Section Review Development Of The Atomic Theory Answers
9. Balancing eBooks and Physical Books Section Review Development Of The Atomic Theory Answers
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Section Review Development Of The Atomic Theory Answers
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Section Review Development Of The Atomic Theory Answers
- Setting Reading Goals Section Review Development Of The Atomic Theory Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Section Review Development Of The Atomic Theory Answers
- Fact-Checking eBook Content of Section Review Development Of The Atomic Theory Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
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 - Interactive and Gamified eBooks

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