



# Selco Baler Operating Manual

**Tobias Bleicker**



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web so the formation of a complex ion is a lewis acid base reaction and notice the equilibrium constant for this  $1.6 \times 10^6$



to the seventh kf is called the formation constant this is a very high value for the equilibrium constant so the equilibrium lies to the right and this a stable complex ion

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web so basing on my understanding of this so far an atom is an element that has the same number of protons and neutrons an ion is an element that has different numbers of protons and electrons an ion is positive when it has more protons than electrons and negative when it has more electrons than ions while isotopes are the same element with

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