

atoms of solid behaves as a diffraction grating.

$$eV = h\nu$$

$$eV = \frac{hc}{\lambda}$$

$$\lambda = \frac{hc}{eV}$$

$V = 10 \text{ kV}$

$$\lambda = \frac{6.6 \times 10^{-34} \times 3 \times 10^8}{1.6 \times 10^{-19} \times 10 \times 1000}$$

$$\lambda = 1.24 \text{ \AA} \text{ wavelength of X-Ray}$$

Crystal $\text{NaCl} = (1.8 - 2) \text{ \AA}$

Visible as UV \rightarrow Reflection, Refraction
 because their λ very high
 as compare to interatomic spacing
 of Ray. its wavelength is very small
 as compare to interatomic
 spacing.

Bragg's Law

X-Ray



Diffraction

Assumptions

- i) Crystal contain various parallel planes



interplanar distance

Crystal
 Regular
 arrangements
 of an
 atoms

one layer of atom
 \rightarrow we consider plane

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M Walker



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