Solubility Curve Worksheet

- Define solubility.
- Look at the graph below. In general, how does temperature affect solubility?

3) Which compound is LEAST soluble at 10 °C?

4) How many grams of KCI can be dissolved in 100g of water at 80°C?

5) How many grams of NaCl can be dissolved in 100g of water at 90°C?

At 40°C, how much KNO₃ can be dissolved in 100g of water?

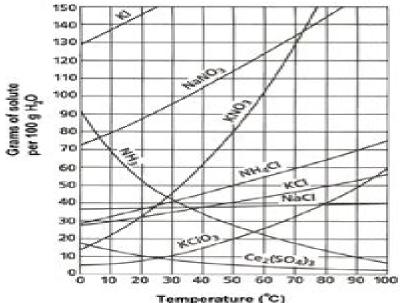
7) Which compound shows the least amount of change in solubility from 0°C-100°C?

8) At 30°C, 90g of NaNO₃ is dissolved in 100g of water. Is this solution saturated or unsaturated?

9) At 60°C, 72g of NH₄Cl is dissolved in 100g of water. Is this solution saturated or unsaturated?

- 10) A saturated solution of KCIO₃ is formed from one hundred grams of water. If the saturated solution is cooled from 90°C to 50°C, how many grams of precipitate are formed?
- 11) A saturated solution of NH₄Cl is formed from one hundred grams of water. If the saturated solution is cooled from 80°C to 40°C, how many grams of precipitate are formed?
- 12) Which compounds show a decrease in solubility from 0°C-100°C?
- 13) Which compound is the most soluble at 10°C?
- 14) Which compound (besides Ce₂(SO₄)₃) is the least soluble at 50°C?
- 15) For each of the following solutions, explain how much of the solute will dissolve and how much will remain undissolved at the bottom of the test tube?

 a) 120 g of KCl in 100 g of water at 80°C
 - b) 130 g of NaNO₃ in 100 g of water at 50°C



Solubility Curve Challenge Answers Key

Fiona Baxter, Liz Dilley

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- 1. Understanding the eBook Solubility Curve Challenge Answers Key
 - The Rise of Digital Reading Solubility Curve Challenge Answers Key
 - o Advantages of eBooks Over Traditional Books
- 2. Identifying Solubility Curve Challenge Answers Key
 - Exploring Different Genres
 - o Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Solubility Curve Challenge Answers Key
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Solubility Curve Challenge Answers Key
 - Personalized Recommendations
 - Solubility Curve Challenge Answers Key User Reviews and Ratings
 - Solubility Curve Challenge Answers Key and Bestseller Lists
- 5. Accessing Solubility Curve Challenge Answers Key Free and Paid eBooks
 - Solubility Curve Challenge Answers Key Public Domain eBooks
 - Solubility Curve Challenge Answers Key eBook Subscription Services
 - Solubility Curve Challenge Answers Key Budget-Friendly Options

- 6. Navigating Solubility Curve Challenge Answers Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Solubility Curve Challenge Answers Key Compatibility with Devices
 - Solubility Curve Challenge Answers Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Solubility Curve Challenge Answers Key
 - Highlighting and Note-Taking Solubility Curve Challenge Answers Key
 - Interactive Elements Solubility Curve Challenge Answers Key
- 8. Staying Engaged with Solubility Curve Challenge Answers Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Solubility Curve Challenge Answers Key
- 9. Balancing eBooks and Physical Books Solubility Curve Challenge Answers Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Solubility Curve Challenge Answers Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Solubility Curve Challenge Answers Key
 - Setting Reading Goals Solubility Curve Challenge Answers Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Solubility Curve Challenge Answers Key
 - Fact-Checking eBook Content of Solubility Curve Challenge Answers Key
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements

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