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Solubility Study Guide

Name	
Teacher:	
Date:	
Period	

References

Chemistry in the Community pp 45-65 Water properties concept map Solubility concept map Solubility, graphing and analysis Solubility Practice: A Curvy Subject Study Solvent Lab

Vocabulary

Understand and apply the definitions or symbols of the following terms.

Colligative properties Distillation Filtration Homogeneous Mixtures Nonpolar Polar Pure substances Solubility Solute
Solvent
Solution
Supersaturated
Saturated
Unsaturated
ö+ and ö-

Targeted Skills

- Describe the structure of water and relate its unusual physical properties to its structure, high boliling point, surface tension, high specific heat.
- Draw the structure of water and relate its structure to the ability of water to act as a universal solvent.
- 3. Compare/contrast the solubilities of various ionic and covalent solutes.
- 4. Apply: Like dissolves like:

Polar solvents dissolve Polar and Ionic solutes

Nonpolar solvents dissolve Nonpolar solutes

- Describe factors that affect the rate of dissolving, such as particle size, stirring, temperature, nature of the solid solutes.
- Interpret information from a solubility curve (graph).

Make sure you remember from past tests?

Atomic structure Significant figures Accuracy & precision

Equations:(extension)

Solution concentration = amount of solute

amount solute + solvent

Test Breakdown

30 Multiple-Choice Problems (max.) - 60%

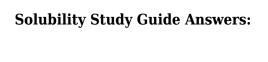
1 Free Response Problem chosen from the Free Response Test Bank - 10%

1 Open-Ended Problem chosen from the Open-Ended Test Bank - 10%.

1 Performance Assessment - 20%

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Ying-Ying Zheng



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