



# Solubility Study Guide

Name \_\_\_\_\_

Teacher: \_\_\_\_\_

Date: \_\_\_\_\_

Period \_\_\_\_\_

## References

Chemistry in the Community pp 45-65  
Water properties concept map  
Solubility concept map  
Solubility, graphing and analysis  
Solubility Practice: A Curvy Subject  
Study Solvent Lab

## Vocabulary

Understand and apply the definitions or symbols of the following terms.

Colligative properties  
Distillation  
Filtration  
Homogeneous  
Mixtures  
Nonpolar  
Polar  
Pure substances  
Solubility

Solute  
Solvent  
Solution  
Supersaturated  
Saturated  
Unsaturated  
 $\delta^+$  and  $\delta^-$

## Targeted Skills

1. Describe the structure of water and relate its unusual physical properties to its structure, high boiling point, surface tension, high specific heat.
2. Draw the structure of water and relate its structure to the ability of water to act as a universal solvent.
3. Compare/contrast the solubilities of various ionic and covalent solutes.
4. Apply: Like dissolves like:  
Polar solvents dissolve Polar and Ionic solutes  
Nonpolar solvents dissolve Nonpolar solutes
5. Describe factors that affect the rate of dissolving, such as particle size, stirring, temperature, nature of the solid solutes.
6. Interpret information from a solubility curve (graph).

## Make sure you remember from past tests?

Atomic structure  
Significant figures  
Accuracy & precision

## Equations: (extension)

Solution concentration =  $\frac{\text{amount of solute}}{\text{amount solute} + \text{solvent}}$

## Test Breakdown

30 Multiple-Choice Problems (max.) – 60%

- 1 Free Response Problem chosen from the Free Response Test Bank – 10%
- 1 Open-Ended Problem chosen from the Open-Ended Test Bank – 10%
- 1 Performance Assessment – 20%

# Solubility Study Guide Answers

**Ying-Ying Zheng**



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