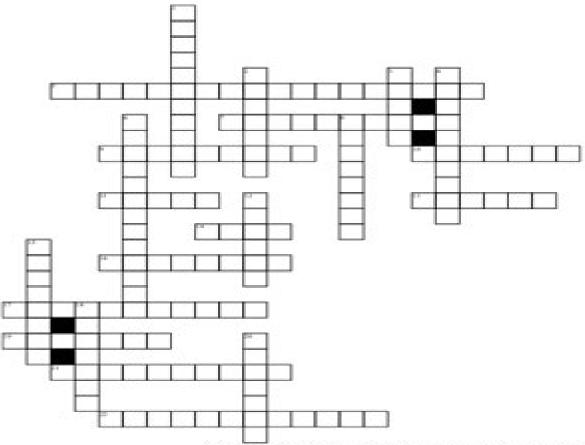
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Atoms and the Periodic Table Vocabulary



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 the elements in Group 2 of the periodic table; they are reactive metals, but are less reactive than alkali metals; their atoms have two electrons in their outer level

- the weighted average of the masses of all he naturally occurring isotopes of an element
- elements that have two properties of both metals and normetals; sometimes referred to as semiconductors.
- 18. the timy, extremely dense, positively charged region in the center of an atom; made up of protons and neutrons.
- 11. a column of elements on the periodic table
- a horizontal row of elements on the periodic table
- 14. the smallest particle into which an element can be divided and still be the same substance

- 16. the negatively charged particles found in all atoms; they are involved in the formation of chemical bonds
- 17. the law that states that the chemical and physical properties of elements are periodic functions of their atomic numbers.
- 19. the elements in Group 17 of the periodic table; they are very reactive normetals, and their atoms have seven electrons in their outer level
- 21. the unreactive elements in Group 18 of the periodic table; their atoms have eight electrons in their outer level
- 22, the number of protons in the nucleus of an atom

Digraviti.

- the elements in Group 1 of the periodic table; they are the most reactive metals, their atoms have one electron in their outer level
- the particles of the nucleus that have no charge

- elements that are shirry and are good conductors of thermal energy and electric current; most of them are mallicable and ductile
- 4. the sum of the protons and neutrons in an
- the regions inside an atom where electrons are likely to be found
- 8. a pure substance composed of two or more elements that are chemically combined
- 12. the positively charged particles of the nucleus; the number of this particle in a nucleus is the atomic number that determines the identity of a substance.
- elements that are dull and are poor conductors of thermal energy and electric current.
- 18, atom that have the same number of protons, but have different numbers of neutrons
- 20. a pure substance that cannot be separated or broken down into simpler substances by physical or chemical means.

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George Graybill

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