

7. At a high temperature, 0.50 mol of HBr was placed in a 1.0 L container and allowed to decompose according to the reaction:



At equilibrium the $[\text{Br}_2]$ was measured to be 0.13 M. What is K_{eq} for this reaction at this temperature?

$$2\text{HBr} \rightleftharpoons \text{H}_2 + \text{Br}_2$$

[I]	0.50	0	0
[C]	-0.26	+0.13	+0.13
[E]	0.24	0.13	0.13

$$K_{\text{eq}} = \frac{[\text{H}_2][\text{Br}_2]}{[\text{HBr}]^2} = \frac{(0.13)^2}{(0.24)^2} = 0.29$$

Answer $K_{\text{eq}} = 0.29$

8. When 1.0 mol of $\text{NH}_3\text{(g)}$ and 0.40 mol of $\text{N}_2\text{(g)}$ are placed in a 5.0 L vessel and allowed to reach equilibrium at a certain temperature, it is found that 0.78 mol of NH_3 is present. The reaction is:



$$\text{initial } [\text{NH}_3] = \frac{1.0 \text{ mol}}{5.0 \text{ L}} = 0.20 \text{ M} \quad \text{initial } [\text{N}_2] = \frac{0.40 \text{ mol}}{5.0 \text{ L}} = 0.080 \text{ M}$$

$$\text{Equil. } [\text{NH}_3] = \frac{0.78 \text{ mol}}{5.0 \text{ L}} = 0.156 \text{ M}$$

$$2\text{NH}_3 \rightleftharpoons 3\text{H}_2 + \text{N}_2$$

[I]	0.20	0	0.080
[C]	-0.044	+0.066	+0.022
[E]	0.156	0.066	0.102

- a) Calculate the equilibrium concentrations of all three species.

$$[\text{NH}_3] = 0.16 \text{ M} \quad [\text{H}_2] = 0.066 \text{ M} \quad [\text{N}_2] = 0.10 \text{ M}$$

- b) Calculate the value of the equilibrium constant at this temperature.

(use unrounded concs, then round to 2 sigs)

$$K_{\text{eq}} = \frac{[\text{H}_2]^3 [\text{N}_2]}{[\text{NH}_3]^2} = \frac{(0.066)^3 (0.102)}{(0.156)^2} = 0.0012$$

Answer 1.2×10^{-3}

- c) How many moles of H_2 are present at equilibrium?

$$0.066 \text{ M} \times 5.0 \text{ L} = 0.33 \text{ mol}$$

Answer 0.33 mol

- d) How many moles of N_2 are present at equilibrium?

$$0.102 \text{ M} \times 5.0 \text{ L} = 0.51 \text{ mol}$$

Answer 0.51 mol

Unit 3 1 Modeling Chemistry

**Tanase Gh. Dobre, José G. Sanchez
Marcano**



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Hazardous Air Pollutant Emissions from Process Units in the Synthetic Organic Chemical Manufacturing Industry, 1993 **Integrated Energy System Modeling and Simulation** Haoyong Chen, Huaguang Yan, Guixiong He, 2025-05-02 This book provides an in depth introduction to the dynamic characteristics of Integrated Energy Systems IES It establishes a unified energy network model elucidating the complementarities between various forms of energy the conversion efficiencies and the interactions and influences on operating costs The book proposes methods for the planning configuration and operational control of IES and develops a simulation system for the integration of multi physical field analog hybrid transient and steady state conditions The established mathematical models and simulation system are utilized to conduct a series of analyses under typical integrated energy scenarios These analyses evaluate the operational performance and economic benefits of IES across different conditions This book serves as a valuable resource for professionals involved in the research planning design consulting research and development and operation of businesses related to integrated energy systems It also serves as an essential reference for educators and college students **Nuclear Science Abstracts**, 1974 **EPA Publications Bibliography Quarterly Abstract Bulletin** United States. Environmental Protection Agency, 2000-04 **Official Gazette** Philippines, 2006 **Springs and Bottled Waters of the World** Philip E. LaMoreaux, Judy T. Tanner, 2012-12-06 Springs and Bottled Waters of the World provides information for the general public legal and scientific professionals and governmental scientists about springs mineral water and thermal waters used for municipal industrial and agricultural water supplies as well as for the rapidly expanding bottled water industry The role of springs is described for ancient civilizations military campaigns and in more recent times for tourism and health spas of the world In addition their source occurrence and methods for development and use are described In the preparation of this book data were obtained from major hydrogeologic data bases and from leading hydrogeologists *ICAR AIEEA UG : All India Entrance Examination for Admission | 10 Full-Length Mock Tests | Physics, Chemistry, and Mathematics/Biology/Agriculture* EduGorilla Prep Experts, 2022-08-03 Best Selling Book for ICAR AIEEA UG Exam with objective type questions as per the latest syllabus given by the National Testing Agency Compare your performance with other students using Smart Answer Sheets in EduGorilla s ICAR AIEEA UG Exam Practice Kit ICAR AIEEA UG Exam Preparation Kit comes with 10 Full length Mock Tests with the best quality content Increase your chances of selection by 14X ICAR AIEEA UG Exam Prep Kit comes with well structured and 100% detailed solutions for all the questions Clear exam with good grades using thoroughly Researched Content by experts **Chemical Engineering** Tanase Gh. Dobre, José G. Sanchez Marcano, 2007-06-18 A description of the use of computer aided modeling and simulation in the development integration and optimization of industrial processes The two authors elucidate the entire procedure step by step from basic mathematical modeling to result interpretation and full scale process performance analysis They further demonstrate similitude comparisons of experimental

results from different systems as a tool for broadening the applicability of the calculation methods Throughout the book adopts a very practical approach addressing actual problems and projects likely to be encountered by the reader as well as fundamentals and solution strategies for complex problems It is thus equally useful for student and professional engineers and chemists involved in industrial process and production plant design construction or upgrading **Resources in Education** ,1980 **Title List of Documents Made Publicly Available** ,1989 Cumulated Index Medicus ,1974

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Symposium on Computer Aided Process Engineering - 14 Ana Paula Barbosa-Póvoa, Henrique Matos,2004-05-14 This book contains papers presented at the 14th European Symposium on Computer Aided Process Engineering ESCAPE 14 The ESCAPE symposia bring together scientists students and engineers from academia and industry who are active in the research and application of Computer Aided Process Engineering The objective of ESCAPE 14 is to highlight the use of computers and information technology tools on five specific themes 1 Product and Process Design 2 Synthesis and Process Integration 3 Process Control and Analysis 4 Manufacturing Process Operations 5 New Challenges in CAPE Provides this year s comprehensive overview of the current state of affairs in the CAPE community Contains reports from the frontiers of science by the field s most respected scientists Special Keynote by Professor Roger Sargent Long Term Achievement CAPE Award winner **Batch Chemical Process Integration** Thokozani Majozi,2010-04-30 Batch Chemical Process Integration Analysis Synthesis and Optimization is an excellent source of information on state of the art mathematical and graphical techniques for analysis synthesis and optimization of batch chemical plants It covers recent techniques in batch process integration with a particular focus on the capabilities of the mathematical techniques There is a section on graphical techniques as well as performance comparison between graphical and mathematical techniques Prior to delving into the intricacies of wastewater minimisation and heat integration in batch processes the book introduces the reader to the basics of scheduling which is aimed at capturing the essence of time A chapter on the synthesis of batch plants to highlight the importance of time in design of batch plants is also presented through a real life case study The book is targeted at undergraduates and postgraduate students researchers in batch process integration practising engineers and technical managers **Char Oil Energy Development** ,1962 *Bulletin of the Chemical Society of Japan* Nihon Kagakkai,1999

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