

## Unit 6: Stoichiometry Review

1. How many moles of hydrogen are required to react with 2.5 moles of oxygen to produce water?



2. How many moles of hydrochloric acid, HCl, will be required to produce 0.40 moles of hydrogen gas by reaction with zinc? The equation is:  $\text{Zn} + 2 \text{HCl} \rightarrow \text{ZnCl}_2 + \text{H}_2$

3. Hydrogen and nitrogen react to form ammonia ( $\text{NH}_3$ ). How many moles of hydrogen will be required to produce 1.70 moles of ammonia,  $\text{NH}_3$ ?

4. Consider the acid-base neutralization reaction between HCl and NaOH. How many moles of HCl are required to completely neutralize 30.0 mL of 0.318 M NaOH?

# Unit 6 Stoichiometry Answers

**Jicheng Xie**



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