Practice Problems for Significant Figures

RULE -1: If the decimal is <u>Present</u>: Find the first non zero on the left, then count all digits to the <u>RIGHT</u>
If the decimal is <u>Absent</u>: Find the first non zero on the right, then count all digits to the <u>LEFT</u>

RULE-2: Every digit in scientific notation is Significant.

RULE-3: Any number that is counted is an EXACT number and has UNLIMITED significant digits.

How many significant figures do the following numbers have?

1.)	1234 m			14)	1000 mm
2)	0.023 L	8)	3.4 x 10 ⁴ g	15)	918.010 L
3)	890 kg	9)	9.0 x 10 ⁻³ cm	16)	0.0001 kg
4)	91010 m	10)	9.010 x 10° g	17)	0.00390 kL
	9010.0 cm	11)	0.00030 hL		8120 cats
5)	218 De 1 Disconstativo la com t. El co	12)	1020010 mg	18)	
6)	1090.0010 g	13)	780. cm	19)	7.991 x 10 ⁻¹⁰ g
7)	0.00120 km	3070	(A. (2) (A. (1) (A. (A	20)	72 students

Solutions to the "How many significant figures"

43.4 51.5 25.4 2) 2 30.2 61.8 73.3 89.2 90.2 101.4 1112 12) 6 13) 3 14) 1 1516 1611 1713 18) unlimited 190.4

Review Game for Sig Figs. http://science.widener.edu/svb/tutorial/sigfigurescsn7.html

Rule for Multiplication & Division = Least number of Sig Figs

Rule for Addition & Subtraction = Least number of decimal places (highest value)

Review Questions for Calculations with Sig Figs. http://www.sciencegeek.net/Chemistry/taters/Unit0Sigfigs.htm

Perform the following calculations and round according to the rule above.



1) 7.6 x 21.9 -			2) 2.15 x 3.1	x 100 -		3) 5.00009 x 0.06 -	
4) 38 / 7 =		5) 500 009 / 17.000 -		6) 500 000 / 5,002 =			
Americani 19179	25799	300.3	418	5 29 412	6,100,000		

1) 334.54 grams + 198 grams = __

2) 34.1 grams / 1.1 mL = _____

3) 2.11 x 10³ joules / 34 seconds =

4) 0.0010 meters - 0.11 m = ___

5) 349 cm + 1.10 cm + 100 cm = _

6) 450 meters / 114 seconds = ___

7) 298.01 kilograms + 34.112 kilograms =

8) 84 m/s x 31.221 s = ___

1) 533 grams 2) 31 g/mL 3) 62 J/s 4) -0.11 m 5) 500 cm 6) .9 m/s 7) 332.12 8) 2600 m

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