# Unit 4 Study Guide: Answer Key 4.1 - 4.11

#### LT 4.1 Types of Elements

LT 4.1 I can identify reactants and products in a chemical equation

$$_2 + O_2 \rightarrow 2H_2O$$

- 1. What are the reactants in the chemical equation above?
  - a. 2H<sub>2</sub> + O<sub>2</sub>
- Explain your answer to #1.
  - Any substance listed on the left hand side of an chemical equation is considered a reactant
- In humans, glucose combines with oxygen to product carbon dioxide and water according to the chemical reaction below:

$$C_6H_{12}O_6 + 6O_2 \rightarrow 6CO_2 + 6H_2O$$

One of your friends who is taking Biology tells you that O<sub>2</sub> is a product because humans make sugar in their bodies. Using your knowledge of chemistry, are they correct or incorrect? Why?

They are incorrect because  $O_2$  is listed on the left hand side of the equation. Therefore  $O_2$  is considered a reactant.

#### LT 4.2 Counting Atoms

\_LT 4.2 I can calculate the number of atoms in a compound

How many atoms of each element are in 4CO<sub>2</sub>?

C: \_\_\_8\_\_\_

2. How many atoms of each element are in 2(NH<sub>4</sub>)<sub>2</sub>CO<sub>3</sub>?

# **Unit 4 Study Guide Chemistry Answers**

John Wilkinson

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