

Acids, Bases and Salts

Reaction of acids with oxides of metals: When a metal oxide reacts with an acid it forms corresponding salt and water. Metallic oxides are basic oxides.

Metal oxide + Dilute acid \longrightarrow Salt + Water



Reactivity of base with non-metallic oxides:

When a non-metallic oxide reacts with a base, it forms salt and water. Non-metallic oxides are acidic in nature.

Non-metallic oxide + Base \longrightarrow Salt + Water



Salts

A salt is a combination of an anion of an acid and a cation of a base.

Examples – KCl, NaNO₃, CaSO₄, etc.

Salts are the ionic compounds which are produced after the neutralization reaction between acid and base. Salts are electrically neutral.

Unit Acids Bases And Salts Answers

Oswaal Editorial Board



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Unveiling the Magic of Words: A Review of "**Unit Acids Bases And Salts Answers**"

In some sort of defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their ability to kindle emotions, provoke contemplation, and ignite transformative change is truly awe-inspiring. Enter the realm of "**Unit Acids Bases And Salts Answers**," a mesmerizing literary masterpiece penned with a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve into the book's central themes, examine its distinctive writing style, and assess its profound affect on the souls of its readers.

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