

Question 10

Mass of NaI crystals	2.0 g
Color of NaI crystals	White
Color of solid recovered	Yellow
Color of filtered solution	Colorless
Mass of dry solid recovered	3.0 g

When students added 2.0 g of NaI crystals to 100. mL of $\text{Pb}(\text{NO}_3)_2(\text{aq})$, a yellow precipitate formed. After the solution was filtered, the yellow solid was dried and weighed. Data from the experiment are shown in the table above. Which of the following claims is best supported by the observations?

- ☐ A A physical change occurred when a new yellow compound was formed.
- ☐ B A physical change occurred when the color of the NaI solid added changed to yellow when mixed with water.
- ☐ C A chemical change occurred when a yellow, insoluble compound with a larger mass than the original NaI formed.
- ☐ D A chemical change occurred when covalent bonds between the yellow solid and water were broken during drying.



Unit 4 Section A Supplement Answers Chemistry

Thomas Griffiths



Unit 4 Section A Supplement Answers Chemistry:

Resources in Education ,1998 **Salters Advanced Chemistry** George Burton,2000-07-31 The texts in the Salters Advanced Chemistry series have been updated to match the specifications for A Level Chemistry from September 2000 This supplement pack is designed to help teachers to use the original editions of the texts until they can be replaced **Bringing Technology Education Into K-8 Classrooms** Edward Britton,Bo De Long-Cotty,Toby Levenson,2005-03-02 Featuring an easy to follow organization and sample pages from major products this resource will help all students become technologically literate Jacket **Final Supplement to the Final Environmental Impact Statements for the Josephine and Jackson-Klamath Sustained Yield Units Ten-year Timber Management Plans** United States. Bureau of Land Management,1985 *Catalog. Supplement - Food and Nutrition Information and Educational Materials Center* Food and Nutrition Information and Educational Materials Center (U.S.),1973 *CHEMISTRY* SILBERBERG,2003 **Export Control Bulletin** United States. Office of Export Control,1972 **Indexes** United States. Environmental Protection Agency,1983 **Nanotechnology Applications for Clean Water** Anita Street,Richard Sustich,Jeremiah Duncan,Nora Savage,2014-05-15 Nanotechnology is already having a dramatic impact on improving water quality and the second edition of Nanotechnology Applications for Clean Water highlights both the challenges and the opportunities for nanotechnology to positively influence this area of environmental protection This book presents detailed information on cutting edge technologies current research and trends that may impact the success and uptake of the applications Recent advances show that many of the current problems with water quality can be addressed using nanosorbents nanocatalysts bioactive nanoparticles nanostructured catalytic membranes and nanoparticle enhanced filtration The book describes these technologies in detail and demonstrates how they can provide clean drinking water in both large scale water treatment plants and in point of use systems In addition the book addresses the societal factors that may affect widespread acceptance of the applications Sections are also featured on carbon nanotube arrays and graphene based sensors for contaminant sensing nanostructured membranes for water purification and multifunctional materials in carbon microspheres for the remediation of chlorinated hydrocarbons Addresses both the technological aspects of delivering clean water supplies and the societal implications that affect take up Details how the technologies are applied in large scale water treatment plants and in point of use systems Highlights challenges and the opportunities for nanotechnology to positively influence this area of environmental protection **Official Gazette of the United States Patent and Trademark Office** ,1996 **Bibliography of Agriculture** ,1974 **Palo Verde Nuclear Generating Station Units 4-5, Construction** ,1979 *Jena Review* ,1980 *Research in Education* ,1973 *Fire Safety for the Aging* United States. Congress. House. Government Operations Committee,1973 **Fire Safety for the Aging** United States. Congress. House. Committee on Government Operations. Special Studies Subcommittee,1973 **Chemical Hydrometallurgy** Alfred Richard Burkin,2001

This book is based on the undergraduate and MSc courses in hydrometallurgy which Professor A R Burkin gave from 1961 until he retired in 1988 It is divided into two sections The first deals with the fundamental chemical and physical principles on which the technology is based In the second processes which are used for the production of individual metals are described in terms of those principles where appropriate

Nuclear Science Abstracts ,1974 **Catalog of Copyright Entries. Third Series** Library of Congress. Copyright Office,1968 Includes Part 1 Number 1 Books and Pamphlets Including Serials and Contributions to Periodicals January June

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