

Name: \_\_\_\_\_

Date: \_\_\_\_\_

### Significant Figures

Determine the number of significant figures in each of these numbers.

	Number	Significant Figures		Number	Significant Figures
1.	357	_____	2.	10000	_____
3.	51015	_____	4.	$6.060 \times 10^{-2}$	_____
5.	0.0007	_____	6.	$4.556 \times 10^{-9}$	_____
7.	5050	_____	8.	5050.0	_____
9.	$6.8 \times 10^3$	_____	10.	0.002110	_____
11.	33.303	_____	12.	170	_____
13.	$15.0 \times 10^{-5}$	_____	14.	0.7007	_____
15.	0.70070	_____	16.	4206	_____
17.	0.02	_____	18.	10.01	_____
19.	0	_____	20.	0.0	_____

# Unit Practice Problems A Significant Figures

**Donald A. McQuarrie, Stanley Gill**



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### **General Chemistry**

Donald A. McQuarrie, Stanley Gill, 2011-06-15 This Fourth Edition of McQuarrie's classic text offers a thorough revision and a quantum leap forward from the previous edition Taking an atoms first approach it promises to be another ground breaking text in the tradition of McQuarrie's many previous works This outstanding new text available in a soft cover edition offers professors a fresh choice

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