

POGIL: Periodic Table Trends

(Adapted from Rush Henrietta CSD)

Why:

The Periodic Table is one of the greatest inventions in the history of man. It allows scientists to predict physical and chemical properties of the elements. Dimitri Mendeleev (a Russian scientist) and Robert Mosley (a British chemist) put together this table in the late 1800's based on properties of the elements known at the time. The trends in these properties as you go across periods and down groups is the subject of this Chem POGIL.

Periodic Table of the Elements

Legend:

- Alkali metals
- Alkaline earth metals
- Transition metals
- Nonmetals
- Metals
- Other metals
- Other nonmetals
- Other metalloids
- Other noble gases
- Other transition metals

Success Criteria:

- Understand the meaning of *atomic radius, reactivity, electronegativity, ionization energy*
- Recognize trends in atomic radius, reactivity, ionization energy and electronegativity as you go across periods and down groups.

Resources:

Reference Table S, Periodic Table, Unit 3 (Chapter 6) Notes, & Chapter 6 in textbook

Prerequisites:

Answer the following questions using the resources above.

- What happens to the number of valence electrons as you go down a group on the periodic table?
- What happens to the number of valence electrons as you go from left to right across a period?
- What happens to the number of energy levels as you go down a group on the periodic table?
- What happens to the number of energy levels as you go from left to right across a period?
- Draw a Lewis structure for a Cl atom.
- Draw a Bohr diagram for the Cl atom. (Draw above!)
- In this lab we will be talking about atomic radius.
 - Look up the atomic radius of Cl on Table S: _____
 - Why is it possible to show the atomic radius using a Bohr diagram but not by using a Lewis structure?
 - On your Bohr diagram draw an arrow to represent the radius of the Cl atom.
- Define *periodic*:
- Define *trend*:

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Table of Contents Pogil Periodic Table Extension Questions Answer Key

1. Understanding the eBook Pogil Periodic Table Extension Questions Answer Key
 - The Rise of Digital Reading Pogil Periodic Table Extension Questions Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Pogil Periodic Table Extension Questions Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Pogil Periodic Table Extension Questions Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Pogil Periodic Table Extension Questions Answer Key
 - Personalized Recommendations
 - Pogil Periodic Table Extension Questions Answer Key User Reviews and Ratings
 - Pogil Periodic Table Extension Questions Answer Key and Bestseller Lists
5. Accessing Pogil Periodic Table Extension Questions Answer Key Free and Paid eBooks
 - Pogil Periodic Table Extension Questions Answer Key Public Domain eBooks
 - Pogil Periodic Table Extension Questions Answer Key eBook Subscription Services
 - Pogil Periodic Table Extension Questions Answer Key Budget-Friendly Options

6. Navigating Pogil Periodic Table Extension Questions Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Pogil Periodic Table Extension Questions Answer Key Compatibility with Devices
 - Pogil Periodic Table Extension Questions Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Pogil Periodic Table Extension Questions Answer Key
 - Highlighting and Note-Taking Pogil Periodic Table Extension Questions Answer Key
 - Interactive Elements Pogil Periodic Table Extension Questions Answer Key
8. Staying Engaged with Pogil Periodic Table Extension Questions Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Pogil Periodic Table Extension Questions Answer Key
9. Balancing eBooks and Physical Books Pogil Periodic Table Extension Questions Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Pogil Periodic Table Extension Questions Answer Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Pogil Periodic Table Extension Questions Answer Key
 - Setting Reading Goals Pogil Periodic Table Extension Questions Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Pogil Periodic Table Extension Questions Answer Key
 - Fact-Checking eBook Content of Pogil Periodic Table Extension Questions Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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web aug 17 2023 the thorax has two major openings the superior thoracic aperture found superiorly and the inferior thoracic aperture located inferiorly the superior thoracic aperture opens towards the neck it is bounded by the bones of the upper thorax manubrium of sternum the first pair of ribs and the body of the vertebra t1

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