

## Naming Acids

What makes an acid unique and how are acids named?

### Why?

A variety of acids are used in foods, industry, and research. Acids are covalently bonded molecules, but when they are put into water they produce ions. One of the ions produced is always  $H^+$ , which immediately combines with a water molecule to form the hydronium ion ( $H_3O^+$ ). The  $H_3O^+$  ion is what defines the acidic properties of a substance. Because of their special classification, acids have a naming system different from ionic or other molecular (covalent) compounds.

### Model 1 – Binary Acids

| Acid   | Name of acid in aqueous solution | Cation (+)* | Anion (–) |
|--------|----------------------------------|-------------|-----------|
| HCl    | Hydrochloric acid                | $H_3O^+$    | $Cl^-$    |
| HBr    | Hydrobromic acid                 | $H_3O^+$    | $Br^-$    |
| $H_2S$ | Hydrosulfuric acid               | $2H_3O^+$   | $S^{2-}$  |
| HF     | Hydrofluoric acid                | $H_3O^+$    | $F^-$     |

\*Hydrogen ions ( $H^+$ ) join with water molecules in solution to form hydronium ions,  $H_3O^+$ .

1. Complete the cation and anion columns of the table in Model 1. Be careful to show the charges on the ions.
2. Why does hydrosulfuric acid contain two hydrogens?

*Two hydrogens with a +1 charge are required to balance the 2– charge of the sulfide anion.*

3. Look at the formulas and names of the binary acids in Model 1.
  - a. What prefix is used at the beginning of the name for all binary acids?

*Binary acids begin with the prefix “hydro-.”*

- b. What suffix is used at the end of the name for all binary acids?

*The names of binary acids end with “-ic,” followed by the word “acid.”*

4. The prefix “bi-” means “two.” Propose a reason that the acids in Model 1 are all referred to as “binary” acids.

*All binary acids have two different elements or atoms in them.*



5. Write a rule for naming binary acids.

*All binary acids begin with the prefix hydro-, which precedes the name of any nonmetal other than hydrogen, in the compound. The name of this nonmetal is modified to end with the suffix -ic.*

*Example: hydro\_\_\_\_\_ic acid.*



# Pogil Naming Acids Worksheet Answers

**Sabine Zange**



## **Pogil Naming Acids Worksheet Answers:**

## Reviewing **Pogil Naming Acids Worksheet Answers**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is really astonishing. Within the pages of "**Pogil Naming Acids Worksheet Answers**," an enthralling opus penned by a highly acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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### **Table of Contents Pogil Naming Acids Worksheet Answers**

1. Understanding the eBook Pogil Naming Acids Worksheet Answers
  - The Rise of Digital Reading Pogil Naming Acids Worksheet Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Pogil Naming Acids Worksheet Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Pogil Naming Acids Worksheet Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Pogil Naming Acids Worksheet Answers
  - Personalized Recommendations
  - Pogil Naming Acids Worksheet Answers User Reviews and Ratings
  - Pogil Naming Acids Worksheet Answers and Bestseller Lists

5. Accessing Pogil Naming Acids Worksheet Answers Free and Paid eBooks
  - Pogil Naming Acids Worksheet Answers Public Domain eBooks
  - Pogil Naming Acids Worksheet Answers eBook Subscription Services
  - Pogil Naming Acids Worksheet Answers Budget-Friendly Options
6. Navigating Pogil Naming Acids Worksheet Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Pogil Naming Acids Worksheet Answers Compatibility with Devices
  - Pogil Naming Acids Worksheet Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Pogil Naming Acids Worksheet Answers
  - Highlighting and Note-Taking Pogil Naming Acids Worksheet Answers
  - Interactive Elements Pogil Naming Acids Worksheet Answers
8. Staying Engaged with Pogil Naming Acids Worksheet Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Pogil Naming Acids Worksheet Answers
9. Balancing eBooks and Physical Books Pogil Naming Acids Worksheet Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Pogil Naming Acids Worksheet Answers
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Pogil Naming Acids Worksheet Answers
  - Setting Reading Goals Pogil Naming Acids Worksheet Answers
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Pogil Naming Acids Worksheet Answers
  - Fact-Checking eBook Content of Pogil Naming Acids Worksheet Answers
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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