

Name _____

Chemistry Worksheet
Naming & Formula Writing (Ionic)

Instructions: Write the formulas &/or the names for the compounds listed below

1. Sodium nitrate _____
2. Calcium carbonate _____
3. Magnesium oxide _____
4. Ammonium sulfide _____
5. Lead (II) sulfate _____
6. Sodium cyanide _____
7. Potassium hydroxide _____
8. Silver chloride _____
9. Iron (III) hydroxide _____
10. Potassium hydroxide _____
11. Tin (IV) perchlorate _____
12. Potassium carbonate _____
13. Silver nitrate _____
14. Sodium iodide _____
15. Ammonium hydroxide _____
16. Potassium iodate _____
17. Lead (IV) oxide _____
18. Ammonium hydroxide _____
19. Barium sulfate _____
20. barium chloride _____
21. Cobalt (II) chloride _____
22. Sodium carbonate _____
23. Calcium oxide _____
24. Lead (II) nitrate _____
25. Tin (II) chloride _____

26. Aluminum chloride _____
27. Iron (III) hydroxide _____
28. Sodium acetate _____
29. calcium hydroxide _____
30. sodium iodate _____
31. Nickel (II) nitrate _____
32. Iron (II) chloride _____
33. Magnesium bromide _____
34. Ammonium nitrate _____
35. Silver bromide _____
36. $\text{Al}(\text{OH})_3$ _____
37. NH_4I _____
38. Li_2CO_3 _____
39. CuSO_4 _____
40. KCN _____
41. $\text{Pb}(\text{ClO})_2$ _____
42. BaS _____
43. ZnSO_4 _____
44. $\text{Pb}(\text{CH}_3\text{COO})_2$ _____
45. $\text{Ca}(\text{NO}_3)_2$ _____
46. $\text{Fe}_2(\text{CO}_3)_3$ _____
47. NH_4IO_3 _____
48. CaCl_2 _____
49. NaF _____
50. $\text{Cu}(\text{NO}_3)_2$ _____

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L Manion

A red circular graphic with a gradient, appearing as a partial circle or a thick curved line, located to the right of the author's name.

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