

Practice Problems for Significant Figures

RULE -1: If the decimal is **Present**: Find the first non zero on the left, then count all digits to the **RIGHT**
 If the decimal is **Absent**: Find the first non zero on the right, then count all digits to the **LEFT**

RULE-2: Every digit in scientific notation is Significant

RULE-3: Any number that is counted is an EXACT number and has UNLIMITED significant digits.

How many significant figures do the following numbers have?

- | | | |
|----------------------|------------------------------------|-------------------------------------|
| 1) 1234 m _____ | 8) 3.4×10^4 g _____ | 14) 1000 mm _____ |
| 2) 0.023 L _____ | 9) 9.0×10^{-3} cm _____ | 15) 918.010 L _____ |
| 3) 890 kg _____ | 10) 9.010×10^{-2} g _____ | 16) 0.0001 kg _____ |
| 4) 91010 m _____ | 11) 0.00030 hL _____ | 17) 0.00390 kL _____ |
| 5) 9010.0 cm _____ | 12) 1020010 mg _____ | 18) 8120 cats _____ |
| 6) 1090.0010 g _____ | 13) 780. cm _____ | 19) 7.991×10^{-10} g _____ |
| 7) 0.00120 km _____ | | 20) 72 students _____ |

Solutions to the "How many significant figures"

- 1) 4 2) 2 3) 2 4) 4 5) 5 6) 8 7) 3 8) 2 9) 2 10) 4
 11) 2 12) 6 13) 3 14) 1 15) 6 16) 1 17) 3 18) unlimited 19) 4 20) unlimited

Review Game for Sig Figs <http://science.widener.edu/syb/tutorial/sigfigurescsn7.htm>

Rule for **Multiplication & Division** = Least number of **Sig Figs**

Rule for **Addition & Subtraction** = Least number of **decimal places** (highest value)

Review Questions for Calculations with Sig Figs <http://www.sciencegeek.net/Chemistry/laters/Unit0Sigfigs.htm>

Perform the following calculations and round according to the rule above.

1) $8.20 + 2 =$	2) $13.59 + 23.25 + 20 =$	3) $42.828 + 67.4629 =$
4) $53.4028 - 14 =$	5) $39.3 - 0.804 =$	6) $91.68 - 19.1 =$
Answers: 1) 10 2) 37 3) 100.299 4) 39 5) 38.5 6) 72.6		
1) $7.6 \times 21.9 =$	2) $2.15 \times 3.1 \times 100 =$	3) $5.00009 \times 0.06 =$
4) $38 / 7 =$	5) $500\ 009 / 17.000 =$	6) $500\ 000 / 5.002 =$
Answers: 1) 170 2) 700 3) 6.3 4) 8 5) 29 412 6) 100 000		

- | | |
|---|--|
| 1) 334.54 grams + 198 grams = _____ | 5) 349 cm + 1.10 cm + 100 cm = _____ |
| 2) 34.1 grams / 1.1 mL = _____ | 6) 450 meters / 114 seconds = _____ |
| 3) 2.11×10^3 joules / 34 seconds = _____ | 7) 298.01 kilograms + 34.112 kilograms = _____ |
| 4) 0.0010 meters - 0.11 m = _____ | 8) 84 m/s x 31.221 s = _____ |

- 1) 533 grams 2) 31 g/mL 3) 62 J/s 4) -0.11 m 5) 500 cm 6) .9 m/s 7) 332.12 8) 2600 m

<http://www.chemfiesta.com>

Unit 08 Practice Problems Chemistry Key Significant Figures

Leo J. Malone, Theodore Dolter



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Mikulecky, 2017-04-17 Take the confusion out of chemistry with hundreds of practice problems Chemistry Workbook For Dummies is your ultimate companion for introductory chemistry at the high school or college level Packed with hundreds of practice problems this workbook gives you the practice you need to internalize the essential concepts that form the foundations of chemistry From matter and molecules to moles and measurements these problems cover the full spectrum of topics you'll see in class and each section includes key concept review and full explanations for every problem to quickly get you on the right track This new third edition includes access to an online test bank where you'll find bonus chapter quizzes to help you test your understanding and pinpoint areas in need of review Whether you're preparing for an exam or seeking a

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