# **Describing Substances Review**

1. Classify each of the pictures below by placing the correct label in the blanks below.

A = Element D = Mixture of Compounds

B = Compound

E = Mixture of Elements and Compounds

C = Mixture of Elements

- A mixture (is/is not) a chemical combining of substances.
- In a compound the (atoms/molecules) are (chemically/physically) combined so that the elements that make up the compound (retain/lose) their identities and (do/do not) take on a new set of properties.
- 4. True or False: A mixture is always made up of a combination of elements.
- In a mixture, the substances (lose/retain) their identities.
- In a mixture the substances involved (can/cannot) be separated by a simple physical process.
- In a compound the elements involved (can/cannot) be separated by a simple physical process because the elements are (physically combined/chemically bonded).
- 8. True or False: An element can be broken down into a simpler substance.

# **Unit 4 Describing Substances Worksheet 1**

**Mary Louise Turgeon** 

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