

Oxidation and Reduction

When copper(II) nitrate is transformed in a chemical reaction?

Q1a)?

Below reactions show the process in copper(II) nitrate transformation in the air. Copper gets oxidized in chemical gases present in the air. Metals react or combine in the presence of air and water. Nitrate(II) ions decompose to give the reaction for decomposition with the use of nitrate(II) ions present in the air. Nitrate(II) ions of other substances are examples of oxidation and reduction. Nitrate(II) ions are called anions. In this reaction, you will observe when it happens in the air and how it reacts in the air.

Model 1 – Redox Reactions

Redox Reactions:



Redox Reactions:



1. What are the redox reactions in Model 1?

Redox reactions and reduction reactions.

2. Explain the redox and reduction reactions in Model 1. What does the reaction in the redox reaction show about the redox reaction? It gives the redox reaction. It gives the redox reaction. It gives the redox reaction.

There is no reaction showing the redox reaction in the redox reaction.

Pogil Lab Oxidation Answer Key

**Los Alamos National Laboratory, G.
Zweig, K.S. Lackner**

A red circular graphic with a gradient, appearing as a partial circle or a thick curved line, located to the right of the text box.

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Analytical Chemistry Juliette Lantz, Renée Cole, The POGIL Project, 2014-12-31 An essential guide to inquiry approach instrumental analysis Analytical Chemistry offers an essential guide to inquiry approach instrumental analysis collection The book focuses on more in depth coverage and information about an inquiry approach This authoritative guide reviews the basic principles and techniques Topics covered include method of standard the microscopic view of electrochemistry calculating cell potentials the BerriLambert atomic and molecular absorption processes vibrational modes mass spectra interpretation and much more *Oxidation and Reduction* ,1992 **Oxidation Numbers of Fractionally Charged Atoms** Los Alamos National Laboratory, G. Zweig, K.S. Lackner, 1983

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