

Chemical Equilibrium - Assignment 1

Questions

- 1. A chemical reaction is at equilibrium when
 - (a) Reactants are completely transformed into products
 - (b) The rates of forward and backward reactions are equal
 - (c) Formation of products is minimized
 - (d) Equal amounts of reactants and products are present
- 2. Which of the following reactions can reach equilibrium in open vessel?
 - (a) NH₄HS(s) == NH₃(g) + H₂S (g)
 - (b) N₂ (g) + 3H₂ (g) ⇒ 2NH₃(g)
 - (c) BaCl₂ (s) + H₂SO₄ (aq) ⇒ BaSO₄ (s) + 2HCl (aq)
 - (d) All of these
- 3. For the reaction, $CO(g) + Cl_2(g) \Longrightarrow COCl_2(g)$ the $\frac{K_p}{K_c}$ is equal to
 - (a) \sqrt{RT}
 - (b) RT
 - (<) Rr
 - (d) 1.0
- 4. The equilibrium constant Kc for the following reaction at 500 K is,

 $A_2(g) + B_2(g) \implies 2AB; K_c = 60$

Select the correct statement/statements:

- (a) Rate constant for forward reaction k_f = Rate constant for backward reaction k_h
- (b) Rate constant for forward reaction k_f > Rate constant for backward reaction k_b
- (c) Rate constant for forward reaction kf < Rate constant for backward reaction kb
- (d) Value of equilibrium constant changes with the use of catalyst.
- 5. What is observed at equilibrium point in the reaction?

 $PCl_{5(g)} \rightleftharpoons PCl_{3(g)} + Cl_{2(g)}$

- (a) Equal volumes of PCI₅, PCI₃ and CI₂ are present.
- (b) Equal masses of PCI₅, PCI₃ and CI₂ are present.
- (c) The concentrations of PCI₅, PCI₃ and CI₂ become constant.
- (d) Reaction comes to a stop.
- For the reaction: H_∗(g) + I_∗(g) → 2HI(g), the equilibrium constant K_∗ changes with

Unit 3 Chemical Equilibrium Assignment 2 Answers

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,1991 Pollutant Fate and Transport in Environmental Multimedia Frank M. Dunnivant, Elliot Anders, 2019-05-07 Bridges the gaps between regulatory engineering and science disciplines in order to comprehensively cover pollutant fate and transport in environmental multimedia This book presents and integrates all aspects of fate and transport chemistry modeling various forms of assessment and the environmental legal framework It approaches each of these topics initially from a conceptual perspective before explaining the concepts in terms of the math necessary to model the problem so that students of all levels can learn and eventually contribute to the advancement of water quality science. The first third of Pollutant Fate and Transport in Environmental Multimedia is dedicated to the relevant aspects of chemistry behind the fate and transport processes It provides relatively simple examples and problems to teach these principles The second third of the book is based on the conceptual derivation and the use of common models to evaluate the importance of model parameters and sensitivity analysis complex equation derivations are given in appendices Computer exercises and available simulators teach and enforce the concepts and logic behind fate and transport modeling The last third of the book is focused on various aspects of assessment toxicology risk benefit cost and life cycle and environmental legislation in the US Europe and China The book closes with a set of laboratory exercises that illustrate chemical and fate and transport concepts covered in the text with example results for most experiments Features more introductory material on past environmental disasters and the continued need to study environmental chemistry and engineering Covers chemical toxicology with various forms of assessment United States European and Chinese regulations and advanced fate and transport modeling and regulatory implications Provides a conceptual and relatively simple mathematical approach to fate and transport modeling yet complex derivations of most equations are given in appendices Integrates the use of numerous software packages pC pH EnviroLab Simulators Water Wastewater and Global Issues and Fate 2016 Contains numerous easy to understand examples and problems along with answers for most end of the chapter problems and simulators for answers to fate and transport questions Includes numerous companion laboratory experiments with EnviroLab Requiring just a basic knowledge of algebra and first year college chemistry to start Pollutant Fate and Transport in Environmental Multimedia is an excellent textbook for upper level undergraduate and graduate faculty and students studying environmental engineering and science

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