

# **PHYS 320B – HOMEWORK 1**

## **Background: your book, [Schaum's Outline of Quantum Mechanics](#)**

Provide the missing concept or mechanism for following physical mechanisms. Support your type of nuclear reaction (or emission, if fission, or annihilation; positron emission, artificial transmutation, fusion, or fission/annihilation)

- 1-1)  ${}^{238}_{92}\text{U} \rightarrow {}^{234}_{90}\text{Th} + \text{_____}$
- 2-1)  ${}^{238}_{92}\text{U} \rightarrow {}^{234}_{92}\text{U} + \text{_____}$
- 3-1)  ${}^{238}_{92}\text{U} \rightarrow {}^{234}_{92}\text{U} + \text{_____}$
- 4-1)  ${}^{238}_{92}\text{U} \rightarrow {}^{234}_{92}\text{U} + \text{_____}$
- 5-1)  ${}^{238}_{92}\text{U} \rightarrow {}^{234}_{92}\text{U} + \text{_____}$
- 6-1)  ${}^{238}_{92}\text{U} \rightarrow {}^{234}_{92}\text{U} + \text{_____}$
- 7-1)  ${}^{238}_{92}\text{U} \rightarrow {}^{234}_{92}\text{U} + \text{_____}$
- 8-1)  ${}^{238}_{92}\text{U} \rightarrow {}^{234}_{92}\text{U} + \text{_____}$
- 9-1)  ${}^{238}_{92}\text{U} \rightarrow {}^{234}_{92}\text{U} + \text{_____}$
- 10-1)  ${}^{238}_{92}\text{U} \rightarrow {}^{234}_{92}\text{U} + \text{_____}$
- 11-1)  ${}^{238}_{92}\text{U} \rightarrow {}^{234}_{92}\text{U} + \text{_____}$
- 12-1)  ${}^{238}_{92}\text{U} \rightarrow {}^{234}_{92}\text{U} + \text{_____}$
- 13-1)  ${}^{238}_{92}\text{U} \rightarrow {}^{234}_{92}\text{U} + \text{_____}$
- 14-1)  ${}^{238}_{92}\text{U} \rightarrow {}^{234}_{92}\text{U} + \text{_____}$
- 15-1)  ${}^{238}_{92}\text{U} \rightarrow {}^{234}_{92}\text{U} + \text{_____}$

Write the balanced nuclear equation for the alpha decay of [Plutonium-239](#)

or [Americium-241](#)

Write the balanced nuclear equation for the alpha decay and gamma decay of [Radium-226](#)

## **General Nuclear Reaction**

- 1-2)  ${}^{238}_{92}\text{U} \rightarrow {}^{234}_{90}\text{Th} + \text{_____}$
- 2-2)  ${}^{238}_{92}\text{U} \rightarrow {}^{234}_{92}\text{U} + \text{_____}$
- 3-2)  ${}^{238}_{92}\text{U} \rightarrow {}^{234}_{92}\text{U} + \text{_____}$
- 4-2)  ${}^{238}_{92}\text{U} \rightarrow {}^{234}_{92}\text{U} + \text{_____}$
- 5-2)  ${}^{238}_{92}\text{U} \rightarrow {}^{234}_{92}\text{U} + \text{_____}$
- 6-2)  ${}^{238}_{92}\text{U} \rightarrow {}^{234}_{92}\text{U} + \text{_____}$
- 7-2)  ${}^{238}_{92}\text{U} \rightarrow {}^{234}_{92}\text{U} + \text{_____}$
- 8-2)  ${}^{238}_{92}\text{U} \rightarrow {}^{234}_{92}\text{U} + \text{_____}$
- 9-2)  ${}^{238}_{92}\text{U} \rightarrow {}^{234}_{92}\text{U} + \text{_____}$
- 10-2)  ${}^{238}_{92}\text{U} \rightarrow {}^{234}_{92}\text{U} + \text{_____}$
- 11-2)  ${}^{238}_{92}\text{U} \rightarrow {}^{234}_{92}\text{U} + \text{_____}$
- 12-2)  ${}^{238}_{92}\text{U} \rightarrow {}^{234}_{92}\text{U} + \text{_____}$
- 13-2)  ${}^{238}_{92}\text{U} \rightarrow {}^{234}_{92}\text{U} + \text{_____}$
- 14-2)  ${}^{238}_{92}\text{U} \rightarrow {}^{234}_{92}\text{U} + \text{_____}$
- 15-2)  ${}^{238}_{92}\text{U} \rightarrow {}^{234}_{92}\text{U} + \text{_____}$

# Unit 16 Nuclear Chemistry Answers

**Arihant Experts**



## Unit 16 Nuclear Chemistry Answers:

*Let's Review Regents: Chemistry--Physical Setting Revised Edition* Barron's Educational Series, Albert S.

Tarendash, 2021-01-05 Barron's Let's Review Regents Chemistry gives students the step by step review and practice they need to prepare for the Regents Chemistry Physical Setting exam. This updated edition is an ideal companion to high school textbooks and covers all Chemistry topics prescribed by the New York State Board of Regents. Let's Review Regents Chemistry covers all high school level Chemistry topics and includes Extensive review of all topics on the test. Extra practice questions with answers. A detailed introduction to the Regents Chemistry course and exam. One actual recently released Regents Chemistry exam with an answer key.

### **Regents Chemistry--Physical Setting Power Pack Revised Edition**

Albert S. Tarendash, 2021-01-05 Barron's two book Regents Chemistry Power Pack provides comprehensive review, actual administered exams and practice questions to help students prepare for the Chemistry Regents exam. This edition includes Regents Exams and Answers. Chemistry Eight actual administered Regents Chemistry exams so students can get familiar with the test. Thorough explanations for all answers. Self analysis charts to help identify strengths and weaknesses. Test taking techniques and strategies. A detailed outline of all major topics tested on this exam. A glossary of important terms to know for test day. Let's Review Regents Chemistry Extensive review of all topics on the test. Extra practice questions with answers. A detailed introduction to the Regents Chemistry course and exam. One actual recently released Regents Chemistry exam with an answer key.

*Basic Concepts of Chemistry* Leo J. Malone, Theodore Dolter, 2008-12-03 Engineers who need to have a better understanding of chemistry will benefit from this accessible book. It places a stronger emphasis on outcomes assessment which is the driving force for many of the new features. Each section focuses on the development and assessment of one or two specific objectives. Within each section a specific objective is included, an anticipatory set to orient the reader, content discussion from established authors and guided practice problems for relevant objectives. These features are followed by a set of independent practice problems. The expanded Making it Real feature showcases topics of current interest relating to the subject at hand such as chemical forensics and more medical related topics. Numerous worked examples in the text now include Analysis and Synthesis sections which allow engineers to explore concepts in greater depth and discuss outside relevance.

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**Radiochemistry and Nuclear Chemistry** Gregory Choppin, Jan-Olov Liljenzin, Jan Rydberg, 2002 Origin of Nuclear Science. Nuclei Isotopes and Isotope Separation. Nuclear Mass and Stability. Unstable Nuclei and Radioactive Decay. Radionuclides in Nature. Absorption of Nuclear Radiation. Radiation Effects on Matter. Detection and Measurement Techniques. Uses of Radioactive Tracers. Cosmic Radiation and Elementary Particles. Nuclear

Structure Energetics of Nuclear Reactions Particle Accelerators Mechanics and Models of Nuclear Reactions Production of Radionuclides The Transuranium Elements Thermonuclear Reactions the Beginning and the Future Radiation Biology and Radiation Protection Principles of Nuclear Power Nuclear Power Reactors Nuclear Fuel Cycle Behavior of Radionuclides in the Environment Appendices Solvent Extraction Separations Answers to Exercises Isotope Chart Periodic Table of the Elements Quantities and Units Fundamental Constants Energy Conversion Factors Element and Nuclide Index Subject Index

**Nuclear Science Abstracts** ,1970      **Chemistry: 1,001 Practice Problems For Dummies (+ Free Online Practice)** Heather Hattori,Richard H. Langley,2014-03-11 Practice makes perfect and helps deepen your understanding of chemistry Every high school requires a course in chemistry and many universities require the course for majors in medicine engineering biology and various other sciences 1001 Chemistry Practice Problems For Dummies provides students of this popular course the chance to practice what they learn in class deepening their understanding of the material and allowing for supplemental explanation of difficult topics 1001 Chemistry Practice Problems For Dummies takes you beyond the instruction and guidance offered in Chemistry For Dummies giving you 1 001 opportunities to practice solving problems from the major topics in chemistry Plus an online component provides you with a collection of chemistry problems presented in multiple choice format to further help you test your skills as you go Gives you a chance to practice and reinforce the skills you learn in chemistry class Helps you refine your understanding of chemistry Practice problems with answer explanations that detail every step of every problem Whether you re studying chemistry at the high school college or graduate level the practice problems in 1001 Chemistry Practice Problems For Dummies range in areas of difficulty and style providing you with the practice help you need to score high at exam time      Revise A2 Chemistry for OCR A Lord Eccles,Wooster,2005-03-29 With short questions at the end of each section that make students stop and think about the topic this work provides tips on common pitfalls and advice on how to tackle different types of exam question and exam preparation It also includes practice exam style questions      **Objective Question Bank in Chemistry** B.K. Sharma,1997      **Principles of Modern Chemistry** David W. Oxtoby,H. Pat Gillis,Laurie J. Butler,2016-01-01 Long considered the standard for honors and high level mainstream general chemistry courses PRINCIPLES OF MODERN CHEMISTRY continues to set the standard as the most modern rigorous and chemically and mathematically accurate text on the market This authoritative text features an atoms first approach and thoroughly revised chapters on Quantum Mechanics and Molecular Structure Chapter 6 Electrochemistry Chapter 17 and Molecular Spectroscopy and Photochemistry Chapter 20 In addition the text utilizes mathematically accurate and artistic atomic and molecular orbital art and is student friendly without compromising its rigor End of chapter study aids focus on only the most important key objectives equations and concepts making it easier for students to locate chapter content while applications to a wide range of disciplines such as biology chemical engineering biochemistry and medicine deepen students understanding of the relevance of chemistry beyond the classroom      The Pearson Guide To Physical

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