Name:	Period:

#### Stoichiometry Unit Study Guide

#### Vocabulary section

Word bank

sermool.

Directions: Use the following words from the word bank to fill in the blanks that best fits the sentence/definitions. Each word is only used ONCE.

Stoichiometry Mole number		Grams Scientific notation	Molar Mass  Law of conservation of mass	Mole ratio  Balanced chemical equation	
1		Record of the Control			
1.	not created	or destroyed. It also m present in the product.	eans that if an element is present an		

- is the unit that measures mass only.
- 4. \_\_\_\_\_\_ is the relationship between the amounts/quantities of substances taking part in a chemical reaction. We use the "train track method" to solve these types of problems.
- The is 1 mole=6.02x10<sup>th</sup> atoms/molecules.
- A \_\_\_\_\_\_ is like the recipe of a chemical reaction and must follow the law of conservation.
- is a way of writing very large or very small numbers to use them more effectively.
- 8. \_\_\_\_ is a conversion factor that relates the amounts in moles of any two substances in a chemical reaction.

Directions: Answer the following questions using dimensional analysis (AKA the train track method). Show all your work and include units!

Find the number of moles in 46.32 grams of H<sub>2</sub>O.

# **Unit Stoichiometry Study Guide Answers**

James E. Brady, John R. Holum

#### **Unit Stoichiometry Study Guide Answers:**

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