

HONORS CHEMISTRY: FINAL EXAM STUDY GUIDE

- The final exam will consist of 80 Multiple Choice questions and 2 open-end questions. The outline provided below describes the major topics and units that will be covered on the multiple choice section of the final exam:

UNIT 6: "Gases and Gas Law Relationships" (~ 10 questions)

- Gases vs. Liquids vs. Solids: Properties and Behavior
- Pressure and Volume: Boyle's Law
- Temperature and Volume: Charles' Law
- The Combined Gas Law
- Moles and The Ideal Gas Law Equation
- Avogadro's Law, Dalton's Law of Partial Pressure, Graham's Law

UNIT 7: "Chemical Bonding and Molecular Geometry" (~ 20 questions)

- Ionic, Covalent and Metallic Bonding: Definitions, Models and Properties
- Using Dot Diagrams to draw Lewis bond structures
- Polar vs. Non-Polar Bonding and Electronegativity Difference
- Isomers and Resonance
- VSEPR Theory, Facts, and Examples
- AB_n Formulas, Bond Length and Bond Angles, Molecule Shapes
- Bond Hybridization Basics
- Intermolecular Forces including Hydrogen Bonding

UNIT 8: "Thermodynamics and Kinetics" (~ 20 questions)

- Definitions of Energy, Enthalpy, Entropy
- Exothermic vs. Endothermic Changes
- Hess's Law Calculations and Interpretation
- Entropy Analysis and Factors Affecting Entropy Change
- Gibbs Free Energy Calculations and Interpretation
- Rate-Influencing Factors and Collision Theory
- Determining Rate Order and Solving Rate Law Expressions
- Reaction Mechanisms
- Catalysts and Energy/Rate Graphical Analysis

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