AECHE - Sem 2

Experiments 35, 39, 40 & 41

Experiment 35

Results

Solute/ Reactant 1	Solvent/ Reactant 2	Initial Temp. °C	Final Temp. °C
NaOH	H ₂ O	18	30
NH₄CI	H₂O	18	17
NaCH ₃ COO	H _z O	18	16
NaCl	H ₂ O	18	16
BaOH ₂	NH ₄ SCN	18	15
Fe	CuSO ₄	18	20

Discussion

- Write equations for the processes or reactions in parts A, B and C and classify each reaction as either exothermic or endothermic
 - a.) NaOH, w Na' + OH' (exothermic)
 - b.) NH₄Cl_(ac) » NH₄*+ Cl* (exothermic)
 - c.) NaCH3COO(sq) ** Na* + CH3COO (exothermic)
 - d.) NaCl (ac) » Na* + Cl (exothermic)
 - e.) BaOH_{2 (aq)} + 2NH₄SCN » Ba(SCN)_{2(aq)} + 2NH_{3 (q)} (endothermic)
 - f.) $CuSO_{4(aq)} + Fe_{(s)} * FeSO_{4(aq)} + Cu²⁺ (aq) (endothermic)$

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