In the list below, circle the properties of acids and box the properties of bases.

Sour taste Feel slippery Contains hydroxide ions

Turns litmus blue Reacts with metals to form Ho Turns litmus red

Bitter taste Contains hydrogen ions Corrosive

Complete the chart below:

Acid-Base Definitions		
Type	Acid Definition	Base Definition
Arrhenius	Donates an H' ion in solution	Donates an OH ion in solution
Bronsted-Lowry	Donates an H* ion	Accepts an H* ion
Lewis	Accepts an electron pair	Donates an electron pair

- What term(s) describe a substance that can act as an acid or a base? Amphoteric or amphiprotic
- 4. Neutralization reactions always produce a SALT and WATER
- If the [H^{*}] is greater than the [OH^{*}], the solution is acidic basic neutral
- 6. If the [OH] is greater than the [H], the solution is acidic basic neutral
- 7. If the [H'] is equal to the [OH'], the solution is acidic basic neutral
- 8. If the pH is 7, the solution is acidic basic neutral
- If the pH is greater than 7, the solution is acidic basic neutral
- 10. If the pH is less than 7, the solution is acidic basic neutral
- 11. How many moles of lithium hydroxide are required to neutralize 19.2 grams of chloric acid?

(19.2 g HClO₂) x (1 mol HClO₂/84 g HClO₂) x (1 mol LiOH/1 mol HClO₂) = 0.228 mol LiOH

12. How many grams of calcium hydroxide are required to neutralize 132.4 grams of nitric acid?

 $Ca(OH)_2 + 2 HNO_3 \rightarrow Ca(NO_3)_2 + 2 H_2O$

(132.4 g HNO₃) x (1 mol HNO₃/63 g HNO₃) x (1 mol Ca(OH)₂/2 mol HNO₃) x (74 g Ca(OH)₂/1 mol Ca(OH)₂) = <u>77.76 g Ca(OH)₂</u>

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