

1. Which number has *the most* significant figures?

300.00 5 sig figs

all zeros in this number are significant as they come at the end of the number with a decimal point

5.78 3 sig figs

0.000019 2 sig figs

32 000 000 2 sig figs

2. Which of the following *correctly presents* the result of the calculations:

$$94.300 \text{ L} + 281.7 \text{ L} = 0.27 \text{ L}$$

375.7 L

the answer should have one digit after decimal, the rule for addition/subtraction

3. The measurement 0.00000284 mg, expressed correctly using exponential notation:

$2.84 \times 10^{-6} \text{ mg}$

The power is negative as the original number is small

4. Which of the following numbers was *correctly rounded* to **three** significant figures?

361.892 to 362 this number is correctly rounded and has 3 sig figs

80.0862 to 81.0 - rounding is incorrect

0.05114 to 511 - rounding is incorrect

9.02438 to 9.024 this number is correctly rounded but has 4 sig figs

5. A pack of Paprika weighs 3.76 oz. Convert this mass into grams? 1 oz = 28.40 g

107 g

$3.76 \text{ oz} \times \frac{28.40 \text{ g}}{1 \text{ oz}} = 106.784 \text{ g}$; the number should be rounded to have 3 sig figs as in a number of oz given

Spring Semester Senior Chemistry Exam Answers

David M. Behrman, Edward J. Behrman



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Measurable Outcomes of Individual Laboratory Work in High School Chemistry Ralph Ellison Horton, 1928

Survival Handbook for the New Chemistry Instructor Diane M. Bunce, Cinzia M. Muzzi, 2004 This book provides an overview of the issues facing new chemistry faculty in preparation for teaching Serving as a reference to answer specific questions new chemistry faculty encounter this book is comparable to sitting down with a colleague in the department and talking through some ideas or gaining some pointers on how to avoid common pitfalls It is the one single place new chemistry faculty can go to find practical information on how to teach and how to prepare for teaching their first course Chapters are written both by established experts in the field and by new professors within their first couple of years of teaching

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