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		Judicia	ry Study Guide	2	
1.	is the right or authority of a court to hear a case.				
2.	the court is to be the first one to hear the case.				
	A is when a trial jury is unable to reach a verdict.				
4.	occurs when a person is prosecuted for the same crime twice.				
5.	A lawyer that is appointed to represent a person who cannot afford one is called a				
6.	What is an indictment?_				
7.	What is a Grand jury?				
8.		is the action	of making a false s	poken statement abo	ut a person.
9.		is the action	of making a false v	vritten statement abo	ut a person.
10.	What is probable cause?				
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24. What is the difference between original and appellate jurisdiction? ______

25. Why should federal courts be able to declare laws unconstitutional?

23. What is presumption of innocence?

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jan 3 2023 a net ionic equation is a chemical equation that shows only the species participating in a chemical reaction and omits the spectator ions spectator ions are ions that

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a net ionic equation is the most accurate representation of the actual chemical process that occurs writing these equations requires a familiarity with solubility rules acid base reactivity

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apr 25 2017 this is understood when you consider the two chemicals appear unchanged before and after the reaction therefore the net ionic equation is 2cl aq 2ag aq

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ions that remain essentially unchanged during a reaction are called spectator ions this means you can ignore them when you write the ionic equation you only need to model how the solid

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may 11 2017 in the net ionic equation any ions that do not participate in the reaction called spectator ions are excluded as a result the net ionic equation shows only the species that

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the first step to writing a net ionic equation is balancing the chemical equation present let's use the reaction between sodium chloride and silver nitrate as an example next we write the

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highlights learning objectives by the end of this section you will be able to derive chemical equations from narrative descriptions of chemical reactions write and balance chemical

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the net ionic equation is the chemical equation that shows only those elements compounds and ions that are directly involved in the chemical reaction notice that in writing the net ionic

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what remains when the spectator ions are removed is called the net ionic equation which represents the actual chemical change occurring between the ionic compounds cl aq

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this net ionic equation indicates that solid silver chloride may be produced from dissolved chloride and silver i ions regardless of the source of these ions these molecular and

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write the overall chemical equation the complete ionic equation and the net ionic equation for the reaction of aqueous silver fluoride with aqueous sodium phosphate to give solid silver

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the complete ionic equation for this reaction is as follows equation 4 13 2ag aq 2f aq 2nh4 aq cr2o72 aq ag2cr2o7 s 2nh4 aq 2f aq because two nh 4 aq

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