

Name _____ Date _____ Class _____

CHAPTER 15 STUDY GUIDE FOR CONTENT MASTERY

Solutions

Section 15.1 What are solutions?

In your textbook, read about the characteristics of solutions.

Use each of the terms below just once to complete the passage.

immiscible	liquid	soluble	solution
insoluble	miscible	solvent	solvent

Air is air! (1) solution of oxygen gas dissolved in nitrogen gas. The oxygen is air is the (2) solvent and nitrogen is the (3) solvent. Because oxygen gas dissolves in a solvent, oxygen gas is air is (4) soluble substances. A substance that does not dissolve is (5) insoluble. (6) liquid solutions are the most common type of solutions. If one liquid is soluble in another liquid, such as acetic acid in water, the two liquids are (7) miscible. However, if one liquid is insoluble in another, the liquids are (8) immiscible.

Read about hydration in aqueous solutions in your textbook.

The diagram shows the hydration of solid sodium chloride to form an aqueous solution. Use the diagram to answer the following questions.



8. Hydration is a solution in which the solvent is water. What is solute?

Solution is the process of surrounding solute particles with solvent particles to form a solution.

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Section 15.1 continued

10. As sodium chloride dissolves in water, what happens to the sodium and chloride ions?

The sodium and chloride ions are separated and surrounded by the water molecules.

11. Explain the attraction of the water molecules around the sodium ions and chloride ions.

Because the sodium ion is positively charged, it attracts the negatively charged portion of the water molecule (the oxygen atom) and repels the positively charged portion of the water molecule (the hydrogen atoms). Because the chloride ion is negatively charged, it attracts the positively charged portion of the water molecule and repels the negatively charged portion.

12. How does the strength of the attraction between water molecules and sodium and chloride ions compare with the strength of the attraction between the sodium ions and chloride ions? How do you know?

The attraction between the water molecules and the sodium and chloride ions is greater than the attraction between the sodium and chloride ions. The greater strength of attraction between the water molecules and the ions is what causes the solution process to occur.

13. List three ways that the rate of solution may be increased.

stirring or shaking the solution, breaking the solute into smaller pieces, and heating the solvent

In your textbook, read about heat of solution, solubility, and factors that affect solubility.

For each statement below, write true or false.

True _____ 14. The overall energy change that occurs when a solution forms is called the heat of solution.

False _____ 15. Solubility is a measure of the maximum amount of solute that dissolves in a given amount of solvent at a specified temperature and pressure.

False _____ 16. Solubility continues to keep as the solubility rate is less than the crystals solubility rate.

True _____ 17. In a saturated solution, solubility and crystallization are in equilibrium.

True _____ 18. Additional solute can be dissolved in an unsaturated solution.

True _____ 19. The solubility of a gas dissolved in a liquid decreases as the temperature of the solution increases.

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Handbook of Research on Advancing Equity and Inclusion Through Educational Technology

Escudeiro, Paula, Escudeiro, Nuno, Bernardes, Oscar, 2023-08-09 Digital technologies play a significant role in the popular imagination about the future of education as they are a prominent aspect of modern education provision and practice across the globe Due to the increased adoption of digital education materials during the COVID 19 pandemic the pedagogical significance of digital technology has been amplified Advancing Equity and Inclusion Through Educational Technology builds upon the available literature in equity and inclusion through educational technology while providing further research opportunities in this dynamic and growing field It provides the opportunity for reflection on this crucial issue by increasing the understanding of the importance of inclusion and equity in the context of educational improvements and providing relevant academic work empirical research findings and an overview of this relevant field of study Covering topics such as sustainable inclusion learning virtual school press programs and generic skills this major reference work is a comprehensive and timely resource for educators and administrators of both K 12 and higher education government officials pre service teachers teacher educators librarians researchers and academicians

Adult Nurse Practitioner Certification Review Guide Sally K. Miller, 2013 Developed especially for practicing nurses preparing for the certification examination offered by the American Nurses Credentialing Center ANCC Adult Nurse Practitioner Review Fifth Edition provides a succinct yet comprehensive review of the core material Intended to serve as an introduction to the testing arena it has been organized to provide test taking strategies techniques and sample test questions Also included is a bibliography for those who need a more in depth discussion of the subject matter in each chapter This new edition contains new information in the content section regarding updated guidelines Additionally new questions reflect these updates Included with the print book is an online access code for JB TestPrep our online interactive testing program to help pass the exam This online program includes all of book s questions and answers practice exams providing feedback and final exams to help prepare readers for the exam JB TestPrep Adult Nurse Practitioner Review also sold separately Don t forget to visit our new nursing certification website www.NursingCertificationSuccess.com

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