

CXC Physics Notes

Section A

SI Base Units

The SI (*Système International d'Unités*) is a globally agreed system of units, with seven base units. Formally agreed by the 11th General Conference on Weights and Measures (CGPM) in 1960, the SI is at the centre of all modern science and technology. The definition and realisation of the base and derived units is an active research topic for meteorologists with more precise methods being introduced as they become available.

There are two classes of units in the SI: **base units and derived units**. The base units provide the reference **used to define all the measurement units of the system**, whilst the derived units are products of base units and are used as measures of derived quantities:

- The seven SI base units, which comprise:
 - The ampere (A) - unit of measurement of electric current
 - The kilogram (kg) - unit of measurement of mass
 - The metre (m) - unit of measurement of length
 - The second (s) - unit of measurement of time
 - The kelvin (K) - unit of measurement of thermodynamic temperature
 - The mole (mol) - unit of measurement of amount of substance
 - The candela (cd) - unit of measurement of luminous intensity

Other units of physics, (joules, pascals etc.) are derived from these base units. It is therefore important to define the derived quantity properly as this will give the answer to a sharp minded physicist on what the derived unit is.

Example, work is defined as force that acts on a body and displaces it from the point of application to the direction of the force. What does this mean exactly?

Force = ma (mass \times acceleration); unit – kg m/s^2

Displacement = s (displacement is **different** from distance, the s is not seconds); unit – m

By the definition, it basically states that work is force acting on a body displacing it in a particular direction. Therefore the SI derived unit of work is the combination of force \times displacement which equates to:

$$\text{kg m/s}^2 \times m = \text{kg m}^2/\text{s}^2$$

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The British National Bibliography Arthur James Wells, 2003

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