

## WebQuest – Solutions

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<http://www.composition.com/quest.asp>

1. What are the two parts of a solution?
2. Define a solute: substance being dissolved.
3. Define solvent: substance doing dissolving.
4. What does a solution concentration measure?

5. Concentrated solutions have a \_\_\_\_\_ (large/sm) amount of solute.
6. Diluted solutions have a \_\_\_\_\_ (small) amount of solute.
7. Describe an unsaturated solution: able to dissolve more at a higher temperature.

8. Describe a saturated solution: contains all solute at given temperature.

9. What happens when you add more solute to an unsaturated solution?  
The solute dissolves in the solvent, so it remains unsaturated.
10. What happens when you add more solute to a saturated solution?

The solution will not be able to dissolve more solute beyond the saturation point.

11. What is water called a "universal solvent"?  
It can dissolve more substances than any other liquid.
12. What makes water so good at dissolving other things?  
Water is a good solvent because of its polarity and ability to form hydrogen bonds.
13. What two factors affect solubility the most?

# Solutions Webquest Answers

**Larry C. Holt, Marcella L. Kysilka**



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