

Answer to Mole Problems**Mole Worksheet KEY**

Name _____

Period _____

Molar Relationships

Directions: Answer the following questions **on a separate sheet of paper**. Set-up all problems using the factor-label method of dimensional analysis and show all your work and units.

1. What is the mass of 7.50 moles of sulfur dioxide (SO_2)? **480g**

$$\frac{7.5\text{mol}}{1} \times \frac{64\text{g}}{1\text{mol}} = 480\text{g}$$

2. How many moles are there in 250.0 grams of sodium phosphate (Na_3PO_4)? **1.52mol**

$$\frac{250\text{g}}{1} \times \frac{1\text{mol}}{164\text{g}} = 1.52\text{mol}$$

3. How many grams of potassium sulfate (K_2SO_4) are there in 25.3 moles? **4402g**

$$\frac{25.3\text{mol}}{1} \times \frac{176\text{g}}{1\text{mol}} = 4402\text{g}$$

4. What is the volume of 0.38 moles of any gas at STP? **8.5L**

$$\frac{0.38\text{mol}}{1} \times \frac{22.4\text{L}}{1\text{mol}} = 8.5\text{L}$$

7. What is the mass of 51 liters of oxygen gas? **73g**

$$\frac{51\text{L}}{1} \times \frac{1\text{mol}}{22.4\text{L}} \times \frac{32\text{g}}{1\text{mol}} = 73\text{g}$$

1. Calculate the number of moles in 32.2-L of NH_3

$$\frac{32.3\text{L}}{1} \times \frac{1\text{mol}}{22.4\text{L}} = 1.44\text{mol}$$

2. Calculate the number of grams in 3.25-mol of AgNO_3

Skill Practice 5 Moles And Formulas Practice Worksheet

Ralph Thomas Myers



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In a global driven by information and connectivity, the power of words has become more evident than ever. They have the capability to inspire, provoke, and ignite change. Such is the essence of the book **Skill Practice 5 Moles And Formulas Practice Worksheet**, a literary masterpiece that delves deep in to the significance of words and their affect our lives. Published by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we will explore the book is key themes, examine its writing style, and analyze its overall impact on readers.

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