

First/Second Semester B.E. Degree Examination, June/July 2011

Engineering Physics

Time: 3 hrs.

Max. Marks:100

- Note: 1. Answer any FIVE full questions, choosing at least two from each part.
 2. Answer all objective type questions only in OMR sheet page 5 of the answer booklet.
 3. Answer to objective type questions on sheets other than OMR will not be valued.
 4. Physical constants : $h = 6.625 \times 10^{-34} \text{ J-S}$, $c = 3 \times 10^8 \text{ ms}^{-1}$, $m_e = 9.1 \times 10^{-31} \text{ kg}$,
 $k = 1.38 \times 10^{-23} \text{ JK}^{-1}$, $e_0 = 8.854 \times 10^{-12} \text{ Fm}^{-1}$

PART - A

1. a. Choose your answers for the following :
 - i) In Compton Effect, the wavelength of the x-rays scattered at an angle $\theta > 0$.
 A) increases B) doesn't change C) decreases D) none of these
 - ii) K_e , K_p and K_α are respective kinetic energy of an e^- , a proton and α - particle of same de-Broglie wavelength, then
 A) $K_e > K_p > K_\alpha$ B) $K_e > K_p < K_\alpha$ C) $K_e < K_p < K_\alpha$ D) $K_e = K_p = K_\alpha$
 - iii) The heavier of the particles has smallest de-Broglie wave length when both of them.
 A) move with same speed B) move with same momentum
 C) move with same kinetic energy D) none of these
 - iv) Matter waves are not electromagnetic waves because
 A) they move with variable velocity B) depend on charge
 C) move with constant velocity D) none of these (04 Marks)
 - b. What are the basic postulates of quantum theory of radiations? Explain how Planck's overcome the drawbacks of Weins law and Rayleigh Jean's law. (06 Marks)
 - c. Define group and phase velocity. Derive the expression for de-Broglie wavelength using group velocity concept. (06 Marks)
 - d. Compute the de Broglie wavelength for a neutron moving with one tenth part of the velocity of light. (04 Marks)
2. a. Choose your answers for the following :
 - i) An electron is moving in a box of length a ; if ψ_1 is the wave function at $x = \frac{a}{4}$ with $n = 1$ and ψ_2 at $x = a$ for $n = 2$, then $\frac{\psi_2}{\psi_1}$ is
 A) $\frac{\sqrt{2}}{a}$ B) $\frac{\sqrt{a}}{2}$ C) 0 D) ∞
 - ii) For a particle in an infinite potential well in its 1st excited state, the probability of finding the particle at the center of box is
 A) 0 B) 0.25 C) 0.5 D) 0.1
 - iii) To become a nuclear constituent, the K.R of e^- must be of the order of
 A) 20 MeV B) 2 MeV C) 20eV D) zero
 - iv) An electron has a speed of 100 m/s accurate to 0.05%. The uncertainty in its position is
 A) 0.01m B) 0.0115m C) 0.024m D) 0.04m (04 Marks)

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