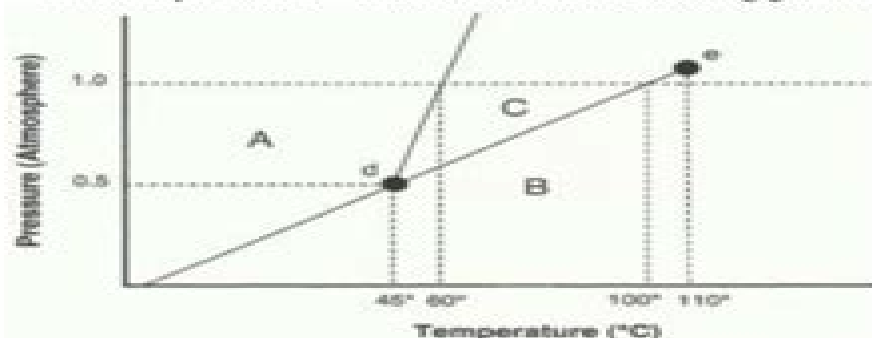


## PHASE DIAGRAM REVIEW WORKSHEET

### Part A – Generic Phase Diagram.

Answer questions 1 – 10 in relation to the following generic phase diagram.



1. Which section represents the solid phase? A
2. What section represents the liquid phase? C
3. What section represents the gas phase? B
4. What letter represents the triple point? d

In your own words, what is the definition of a triple point?

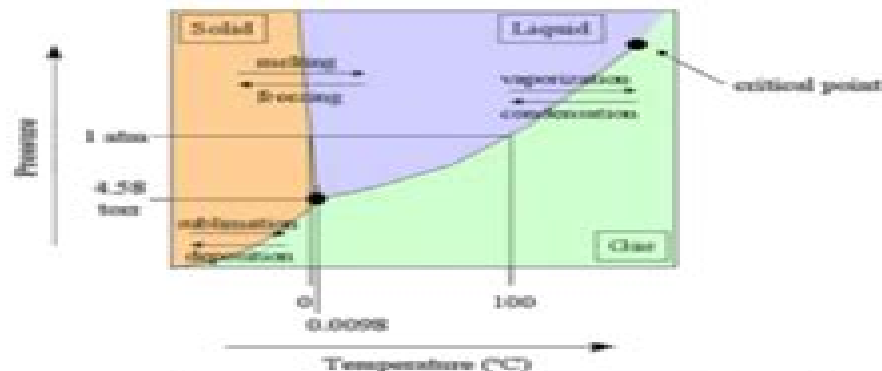
answers will vary – s, l, g exist at equilibrium

5. What is this substance's normal melting point, at 1 atmosphere of pressure? 60°C
6. What is this substance's normal boiling point, at 1 atmosphere of pressure? 100°C

7. Above what temperature is it impossible to liquefy this substance, no matter what the pressure? 110°C
8. At what temperature and pressure do all three phases coexist? 0.5 atm, 45°C
9. At a constant temperature, what would you do to cause this substance to change from the liquid phase to the solid phase? increase the pressure
10. What does sublimation mean? to go from a solid to a gas directly

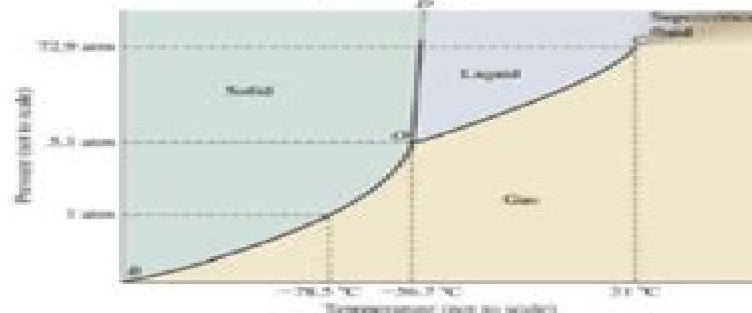
### Part B – Phase Diagram for Water.

11. At a pressure of 1 atmosphere, what is the normal freezing point of water? 0°C
12. What is the normal boiling point of water, at one atmosphere of water? 100°C
13. People in Albuquerque live approximately 5,500 feet above sea level, which means the normal atmospheric pressure is less than 1 atm. In Albuquerque, will water freeze at a lower temperature or a higher temperature than at 1 atmosphere? higher Will water boil at a higher or lower temperature, than at 1 atmosphere? lower



### Part C – Phase Diagram for Carbon Dioxide.

14. At 1 atmosphere and room temperature (25°C), would you expect solid carbon dioxide to melt to the liquid phase, or sublime to the gas phase? sublime to gas
15. Some industrial processes require carbon dioxide. The carbon dioxide is stored on-site in large tanks as liquid carbon dioxide. Assuming we lived at sea level (1 atm), how could carbon dioxide be liquefied? Increase the pressure



# Phase Diagram Part C Answer Key

**Harry Chandler**



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**Guide to the Phase Diagrams of the Fluoride Systems** John Ettore Ricci,1962      **Theory and Practice of Physical Pharmacy - E-Book** Gaurav Jain,2011-06-23 A core subject in pharmaceutics physical pharmacy is taught in the initial semesters of B Pharm The methodical knowledge of the subject is required and is essential to understand the principles pertaining to design and development of drug and drug products Theory and Practice of Physical Pharmacy is unique as it fulfils the twin requirements of physical pharmacy students the authentic text on theoretical concepts and its application including illustrative exercises in the form of practicals Covers all the topics included in various existing syllabi of physical pharmacy Provides an integrated understanding of theory and practical applications associated with physicochemical concepts Explore the latest developments in the field of pharmaceutics Reviews the relevance of physicochemical principles in the design of dosage form Ensures proper recapitulation through sufficient end of chapter questions Provides valuable learning tool in the form of multiple choice questions Multiple choice questions section especially useful for GPAT aspirants

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review from various areas of science We believe that readers will find this Special Issue informative as to the recent advancements of gel research and the broad background of gel science

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CERAMIC MATERIALS AND PROCESSING - PHASE EQUILIBRIA - PHYSICAL MODELLING IN ELECTRON MICROSCOPY Prof. Dr. ZİYA ENGİN ERKMEN, 2019-09-15 There were two main driving forces in my decision for preparing a question answer book covering all the courses given by myself during the past 10 years in my academic career first argument is that there exists a good amount of original questions in the exams and their corresponding answers next my expectation is that probably fewer time would be spent in preparing such a book where the questions and answers are all ready prepared thereupon In this country most of the undergraduate students do not necessarily work on the course they attend in the same day instead they prefer to start preparing their exams 2 or 3 days before In these circumstances last minute students may usually prefer working on the passed questions for the tomorrow's exam However this method of preparation may lead unwanted consequences such that students mostly do not find same questions unconsciously learnt by heart in the exams In order to increase working efficiency and consequently reach to the maximum performance in exams will only be possible if students do not miss any lectures given by staff in addition to be an active participant during the lecture and daily work on the given homeproblems are the basic requirements for a full success In concluding this book will give a chance to the well prepared students to make a quick rehearsal before the exams for obtaining best results after the exams *NBS Publications Newsletter*, 1980 A newsletter for librarians documentalists and science information specialists

**Physical Metallurgy** R.W. Cahn, P. Haasen, 1996-02-09 This is the fourth edition of a work which first appeared in 1965 The first edition had approximately one thousand pages in a

single volume This latest volume has almost three thousand pages in 3 volumes which is a fair measure of the pace at which the discipline of physical metallurgy has grown in the intervening 30 years Almost all the topics previously treated are still in evidence in this version which is approximately 50% bigger than the previous edition All the chapters have been either totally rewritten by new authors or thoroughly revised and expanded either by the third edition authors alone or jointly with new co authors Three chapters on new topics have been added dealing with dry corrosion oxidation and protection of metal surfaces the dislocation theory of the mechanical behavior of intermetallic compounds and most novel a chapter on polymer science for metallurgists which analyses the conceptual mismatch between metallurgists and polymer scientists way of looking at materials Special care has been taken throughout all chapters to incorporate the latest experimental research results and theoretical insights Several thousand citations to the research and review literature are included in this edition There is a very detailed subject index as well as a comprehensive author index The original version of this book has long been regarded as the standard text in physical metallurgy and this thoroughly rewritten and updated version will retain this status

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