

Answers			Marks	Examiner's tips
1 a)	(i)	2	10	2.35
	н)	1	10	
	Ш)	0	1	It is easier to calculate the order with respect to B first in this example.
b)) k=	rate D[E]	3	Make sure you can rearrange formulae. The units are worth a mark so do not leave them out
	or - 1	$\frac{8 \cdot 36 \times 10^{-6}}{(0.84)^{2}(1.16)}$ $0.2 \times 10^{-3} \text{ mol}^{-2} \text{ dim}^{6} \text{ s}^{-1}$		Answers from 1.02 × 10 ⁻³ to 1.05 × 10 ⁻³ are acceptable to take account of rounding up in the numbers during the calculation.
2 a)	i)	Expt 2 2.68 × 10 ⁻⁴ Expt 3 10.72 × 10 ⁻⁴	3	Fewer than 2 dp here will be penalised. This gives 3 significant figs in answer.
		Expt 4 2.08 × 10 ⁻³		Accept 10.7 × 10 ⁻⁴
	н)	$k = \frac{\text{rate}}{[X]^2}$ or $\frac{2.68 \times 10^{-6}}{(1.20 \times 10^{-5})^2}$	3	Units are worth a mark so don't leave them out
		= 186 mol ⁻¹ dm ³ s ⁻¹		In this case candidates often do part iii) to calculate k before part i).
b	inc	eases exponentially, i.e.	1	This is quite hard so a straight line like this was allowed:

Nelson Thornes A2 Chemistry Answers Chapter 6

Michael Robinson, Mike Taylor

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A-level Chemistry E. N. Ramsden, 2000 Each topic is treated from the beginning without assuming prior knowledge Each chapter starts with an opening section covering an application These help students to understand the relevance of the topic they are motivational and they make the text more accessible to the majority of students Concept Maps have been added which together with Summaries throughout aid understanding of main ideas and connections between topics Margin points highlight key points making the text more accessible for learning and revision Checkpoints in each chapter test students understanding and support their private study Essential A2 Chemistry for OCR Janet Renshaw, Ted Lister, 2004 Essential A2 Chemistry for OCR provides clear progression with challenging material for in depth learning and understanding Written by the best selling authors of New Understanding Chemistry these texts have been written in simple easy to understand language and each double page spread is designed in a contemporary manner Fully networkable and editable Teacher Support CD ROMs are also available for this series containing worksheets marking schemes and practical help Advanced Chemistry for You Lawrie Ryan, 2000 Advanced Chemistry for You has been carefully designed to be interesting and motivating to the AS A2 student with features that make it highly supportive of individual learning Written by an experienced author team with the same straightforward approach as the successful New for You GCSE series

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The School Science Review, 2002

Chemistry of the Environment E. N. Ramsden,1996 Written for students undertaking Environmental Chemistry options Concise student friendly and well illustrated with diagrams tables and charts Equally suitable for use as stand alone texts or as ancillary texts to any core chemistry text

Maths for Advanced Chemistry Michael Robinson, Mike Taylor, 2002 Written by teachers this text is useful for both classroom work and homework exercises

New Understanding Chemistry for Advanced Level Third Edition Ted Lister, Janet Renshaw, 2000 Matches the specifications of the Awarding Bodies AQA NEAB AEB OCR and Edexcel This accessible text includes frequent hints questions and examination questions providing support and facilitating study at home It features photographs and comprehensive illustrations with 3D chemical structures

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