



**PHYSICAL SCIENCES (CHEMISTRY)
GRADE 11 TERM 3 FORMAL EXPERIMENT - PART 2**

TIME: 1 HOUR

**VERIFICATION OF BOYLE'S LAW
INDIVIDUAL WORK (should be done under strict supervision).**

1. Write a scientific report which must include the following:

1.1 Aim

(1)

1.2 Apparatus and method

(5)

2. Observations and Recording of results (Copy the table below onto your book or sheet of paper and transfer the results you recorded when conducting the experiment). Calculate pV for each set of results and record your answers in the table.

Volume (V) cm^3	Pressure (p) kPa		pV (J)

(6)

3. Analysis of the results:

3.1 Identify the following variables for the experiment:

- Independent
- Dependent
- Controlled

(3)

3.2 Use the graph paper provided by your teacher to draw graphs of V versus p and V versus $\frac{1}{p}$.

(18)

4. Conclusion

4.1 Examine the results in your table as well as the two graphs you have drawn, what happens to the volume of the gas in the tube as the pressure is decreased.

(1)

4.2 Write down a conclusion for this experiment.

(3)

5

Questions and discussion:

5.1 Explain why you have to wait for a few minutes before you can take readings.

(2)

5.2 Explain the relationship between pressure and volume in terms of the kinetic theory of gases.

(4)

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Table of Contents Physics Paper Grade 1mpumalanga

1. Understanding the eBook Physics Paper Grade 1mpumalanga
 - The Rise of Digital Reading Physics Paper Grade 1mpumalanga
 - Advantages of eBooks Over Traditional Books
2. Identifying Physics Paper Grade 1mpumalanga
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physics Paper Grade 1mpumalanga
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physics Paper Grade 1mpumalanga
 - Personalized Recommendations
 - Physics Paper Grade 1mpumalanga User Reviews and Ratings
 - Physics Paper Grade 1mpumalanga and Bestseller Lists

5. Accessing Physics Paper Grade 1mpumalanga Free and Paid eBooks
 - Physics Paper Grade 1mpumalanga Public Domain eBooks
 - Physics Paper Grade 1mpumalanga eBook Subscription Services
 - Physics Paper Grade 1mpumalanga Budget-Friendly Options
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 - ePub, PDF, MOBI, and More
 - Physics Paper Grade 1mpumalanga Compatibility with Devices
 - Physics Paper Grade 1mpumalanga Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Physics Paper Grade 1mpumalanga
 - Highlighting and Note-Taking Physics Paper Grade 1mpumalanga
 - Interactive Elements Physics Paper Grade 1mpumalanga
8. Staying Engaged with Physics Paper Grade 1mpumalanga
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physics Paper Grade 1mpumalanga
9. Balancing eBooks and Physical Books Physics Paper Grade 1mpumalanga
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physics Paper Grade 1mpumalanga
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Physics Paper Grade 1mpumalanga
 - Setting Reading Goals Physics Paper Grade 1mpumalanga
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Physics Paper Grade 1mpumalanga
 - Fact-Checking eBook Content of Physics Paper Grade 1mpumalanga
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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