Matter and Change, pp. 1-4

	5.	40
2.	.5.	40
3.	7.	400
4.	1.0	100
We will be the second of the s		100

2. mass and volume

group or family

11. homogeneous

products

13. metal

solid state

chemical properties

10. attorns

17. compound.

18. nonmetal

19. mixture

20. pune substance

21. busic

22. metalloid

reactants: carbon and oxygen; product: carbon dioxide

 reactant: mercury(II) oxide; products: mercury and oxygen

25. 6

26.

27. 8

28. **

30. b

31. 1

32.

33. "

24 6

26 physical change

16 chemical change

36. physical change

38. physical change

chemical change

to physical change

40. In a solid, particles are packed together in relatively fixed positions. The particles vibrate about a fixed point. In i liquid, particles are close together but can flow a 'ound one another. In a gas, particles are at gre it distances from one another, compared to tile particles of liquids and solids.

41. The composition of a pure substance is the same throughout and does not vary from sample to sample. Pure substances include elements and compounds. A mixture contains more than one substance and can vary in composition from one sample to another and s ometimes within different parts of the same sample.

42. Although energy can be absorbed or released in a

change, it is not destroyed or created. It simply takes a different form.

 Homogeneous mixtures are uniform in composition; heterogeneous mixtures are not.

 An element cannot be broken down, but water can be decomposed into hydrogen and oxygen by passing an electrical current through it (electrolysis).

Measurements and Calculations, pp. 5-8

	a	5. c
2.	C C	6. €
3.	C	7. a
4.		8. a
4. 9.	timor	

10. mass

11. density

12. onergy

13. length or distance

14. volume

15. area

16. qualitative

17. quantitative

18. qualitative

quantitative
3.00 X 10⁴ km/s.

3.00 :
three

22. 0.026 g or 2.6 X 10-2 g

23. 2.5 X 10" L

24. quantity

s derived units

26. conversion factor

27. 0.0432 kg or 4.32 X 10⁻¹kg

28. 5400 mL or 5.4 X 103 mL

20 300 K

no. 1.05 X 105 J

31. 3.51 X 10°cm

32. A hypothesis is a testable statement that can be used to make predictions and to carry out further experiments. A theory is a broad generalization that explains a body of facts or phenomena.

 Mass is a measure of the quantity of matter.
Weight is a measure of the gravitational pull on matter. Mass does not depend on gravitational attraction.

34. Models are visual, verbal, or mathematical representations. They are used to explain how phenomena occur or how data or events are related.

 The graph of two quantities that are directly proportional is a straight line through the origin.

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SJ Ball

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