

Answer following



- 1. Substrate and temperature: 20°C 30°C 40°C
- 2. A graph of initial velocity: 100 mg/ml 200 mg/ml 300 mg/ml
- 3. A graph of initial velocity: 100 mg/ml 200 mg/ml 300 mg/ml

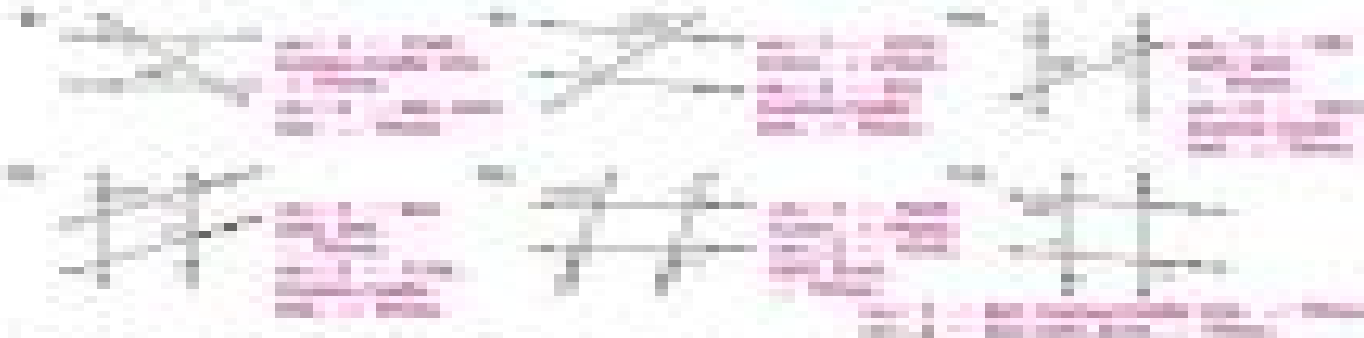
Calculate all values of K_m and V_{max} for the following reaction. (Note: K_m is the Michaelis-Menten constant, V_{max} is the maximum velocity.)

- 1. $K_m = 100\text{ mg/ml}$ $V_{max} = 100\text{ mg/ml}$
- 2. $K_m = 100\text{ mg/ml}$ $V_{max} = 200\text{ mg/ml}$
- 3. $K_m = 100\text{ mg/ml}$ $V_{max} = 300\text{ mg/ml}$
- 4. $K_m = 100\text{ mg/ml}$ $V_{max} = 400\text{ mg/ml}$



Answer: 100

Graph the following data. (Note: K_m is the Michaelis-Menten constant, V_{max} is the maximum velocity.)



Graph the following data. (Note: K_m is the Michaelis-Menten constant, V_{max} is the maximum velocity.)



Michaelis-Menten equation: $V = \frac{V_{max} [S]}{K_m + [S]}$

Prentice Hall Mathematics Geometry Chapter 6 Answers

John K. Beem



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IBM SPSS for Introductory Statistics Karen C. Barrett, Nancy L. Leech, Gene W. Gloeckner, George A.

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