

(English Version)

- Note :
- (i) You must write the subject-code/paper-code 053/B in the box provided on the title page of your answer-book.
  - (ii) Make sure that the answer-book contains 32 pages (including title page) and are properly serialised as soon as you receive it.
  - (iii) Question/s attempted after leaving blank page/s in the answer-book would not be evaluated.
  - (iv) Log tables may be asked for if needed.
  - (v) Use of simple calculator is allowed.
  - (vi) Marks of each question are indicated against it.
  - (vii) Questions no. 1 has 28 sub-parts and each carries one mark.
  - (viii) Questions no. 2 to 11 carry two marks. All questions are compulsory.
  - (ix) Questions no. 12 to 15 carry three marks. All questions are compulsory.
  - (x) Questions no. 16 and 17 carry five marks. All questions are compulsory. There is internal choice in them.



1. (i) Which of the following is hexadentate ligand ?
- (a) Diene
  - (b)  $\text{CN}^-$
  - (c) en
  - (d) ☒ EDTA
- (ii) What is oxidation state of Fe in  $\text{K}_3[\text{Fe}(\text{CN})_6]$  ?
- (a) +2
  - (b) ☒ +3
  - (c) +4
  - (d) +1
- (iii) The correct IUPAC name of  $[\text{Pt}(\text{NH}_3)_2\text{Cl}_2]$  is :
- ☒ (a) Diammine dichlorido platinum (II)
  - (b) Diammine dichlorido platinum (IV)
  - (c) Diammine dichlorido platinum (0)
  - (d) Diammine chloride platinum (IV)
- (iv) In  $[\text{Cr}(\text{NH}_3)_4\text{Cl}_2]$  the ligands are :
- (a)  $\text{NH}_3$  only
  - (b)  $\text{Cl}^{-1}$  only
  - (c) ☒ Both  $\text{NH}_3$  and  $\text{Cl}^{-1}$
  - (d) Cr,  $\text{NH}_3$ ,  $\text{Cl}^{-1}$
- (v) Which of the following reagent is used as a refrigerant ?
- ☒ (a)  $\text{CCl}_4$
  - (b)  $\text{CH}_2\text{Cl}_2$
  - (c)  $\text{CF}_2\text{Cl}_2$
  - (d)  $\text{CF}_4$
- (vi) Phenol on distillation with Zinc dust gives :
- ☒ (a) Benzene
  - (b) Benzaldehyde
  - (c) Benzoic Acid
  - (d) Benzophenone
- (vii) Which of the following is strongest acid ?
- ☒ (a)  $\text{CH}_3\text{COOH}$
  - (b)  $\text{CH}_3\text{CHClCOOH}$
  - (c)  $\text{CH}_3\text{CH}_2\text{COOH}$
  - (d)  $\text{CH}_2\text{ClCH}_2\text{COOH}$



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