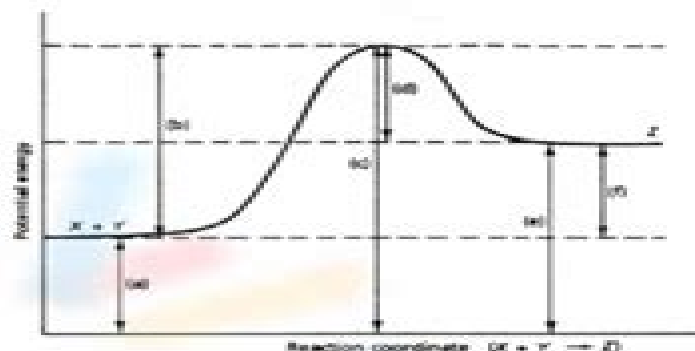
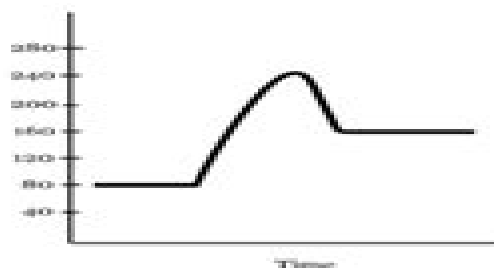


Potential Energy Diagram Worksheet **ANSWERS**



1. Which of the letters a-f in the diagram represents the potential energy of the products? **e**
2. Which letter indicates the potential energy of the activated complex? **c**
3. Which letter indicates the potential energy of the reactants? **a**
4. Which letter indicates the activation energy? **b**
5. Which letter indicates the heat of reaction? **f**
6. Is the reaction exothermic or endothermic? **endo**

7. Which letter indicates the activation energy of the reverse reaction? **d**
8. Which letter indicates the heat of reaction of the reverse reaction? **f**
9. Is the reverse reaction exothermic or endothermic? **exo**



(endothermic or exothermic).

1. The PE of the reactants of the forward reaction is about **80** kilojoules.
2. The PE of the products of the forward reaction is about **160** kilojoules.
3. The PE of the activated complex of the forward reaction is about **240** kilojoules.
4. The activation energy of the forward reaction is about **160** kilojoules.
5. The heat of reaction (ΔH) of the forward reaction is about **+80** kilojoules.
6. The forward reaction is **endothermic**
7. The PE of the reactants of the reverse reaction is about **160** kilojoules.
8. The PE of the products of the reverse reaction is about **80** kilojoules.
9. The PE of the activated complex of the reverse reaction is about **240** kilojoules.
10. The activation energy of the reverse reaction is about **80** kilojoules.
11. The heat of reaction (ΔH) of the reverse reaction is about **-80** kilojoules.
12. The reverse reaction is **exothermic** (endothermic or exothermic).

Potential Energy Diagram Worksheet

Gurukul

A decorative red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the Gurukul text.

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Pm Science Practice P5/6 , American Journal of Physics ,1998 **Chemistry Homework** Frank Schaffer Publications, Joan DiStasio, 1996-03 Includes the periodic table writing formulas balancing equations stoichiometry problems and more

Solutions Manual for Principles of Physical Chemistry, 3rd Edition Hans Kuhn, David H. Waldeck, Horst-Dieter Försterling, 2024-10-29 This is a Solutions Manual to Accompany with solutions to the exercises in the main volume of Principles of Physical Chemistry Third Edition This book provides a unique approach to introduce undergraduate students to the concepts and methods of physical chemistry which are the foundational principles of Chemistry The book introduces the student to the principles underlying the essential sub fields of quantum mechanics atomic and molecular structure atomic and molecular spectroscopy statistical thermodynamics classical thermodynamics solutions and equilibria electrochemistry kinetics and reaction dynamics macromolecules and organized molecular assemblies Importantly the book develops and applies these principles to supramolecular assemblies and supramolecular machines with many examples from biology and nanoscience In this way the book helps the student to see the frontier of modern physical chemistry developments The book begins with a discussion of wave particle duality and proceeds systematically to more complex chemical systems in order to relate the story of physical chemistry in an intellectually coherent manner The topics are organized to correspond with those typically given in each of a two course semester sequence The first 13 chapters present quantum mechanics and spectroscopy to describe and predict the structure of matter atoms molecules and solids

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Waldeck, Horst-Dieter Försterling, 2024-10-25 This is a Solutions Manual to Accompany with solutions to the exercises in the main volume of Principles of Physical Chemistry Third Edition This book provides a unique approach to introduce undergraduate students to the concepts and methods of physical chemistry which are the foundational principles of Chemistry The book introduces the student to the principles underlying the essential sub fields of quantum mechanics atomic and molecular structure atomic and molecular spectroscopy statistical thermodynamics classical thermodynamics solutions and equilibria electrochemistry kinetics and reaction dynamics macromolecules and organized molecular assemblies Importantly the book develops and applies these principles to supramolecular assemblies and supramolecular machines with many examples from biology and nanoscience In this way the book helps the student to see the frontier of modern physical chemistry developments The book begins with a discussion of wave particle duality and proceeds systematically to more complex chemical systems in order to relate the story of physical chemistry in an intellectually coherent manner The topics are organized to correspond with those typically given in each of a two course semester sequence The first 13 chapters present quantum mechanics and spectroscopy to describe and predict the structure of matter atoms molecules and solids Chapters 14 to 29 present statistical thermodynamics and kinetics and applies their principles to understanding equilibria chemical transformations macromolecular properties and supramolecular machines Each chapter of the book begins with a simplified view of a topic and evolves to more rigorous description in order to provide the student and instructor flexibility to choose the level of rigor and detail that suits them best The textbook treats important new directions in physical chemistry research including chapters on macromolecules principles of interfaces and films for organizing matter and supramolecular machines as well as including discussions of modern nanoscience spectroscopy and reaction dynamics throughout the text

A Training Manual in Conducting a Workshop in the Design, Construction, Maintenance and Repair of Hydrams, 1981

Molecular Quantum Mechanics Peter W. Atkins, Ronald S. Friedman, 2011 This text unravels those fundamental physical principles which explain how all matter behaves It takes us from the foundations of quantum mechanics through quantum models of atomic molecular and electronic structure and on to discussions of spectroscopy and the electronic and magnetic properties of molecules

STEM Road Map Carla C. Johnson, Erin E. Peters-Burton, Tamara J. Moore, 2015-07-03 STEM

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the ICT Age Minu Gupta Bhowon, Sabina Jhaumeer-Laulloo, Henri Li Kam Wah, Ponnadurai Ramasami, 2009-07-21 th th The 20 International Conference on Chemical Education 20 ICCE which had rd th Chemistry in the ICT Age as the theme was held from 3 to 8 August 2008 at Le Meridien Hotel Pointe aux Piments in Mauritius With more than 200 participants from 40 countries the conference featured 140 oral and 50 poster presentations th Participants of the 20 ICCE were invited to submit full papers and the latter were subjected to peer review The selected accepted papers are collected in this book of proceedings This book of proceedings encloses 39 presentations covering topics ranging from fundamental to applied chemistry such as Arts and Chemistry Education Biochemistry and Biotechnology Chemical Education for Development Chemistry at Secondary Level Chemistry at Tertiary Level Chemistry Teacher Education Chemistry and Society Chemistry Olympiad Context Oriented Chemistry ICT and Chemistry Education Green Chemistry Micro Scale Chemistry Modern Technologies in Chemistry Education Network for Chemistry and Chemical Engineering Education Public Understanding of Chemistry Research in Chemistry Education and Science Education at Elementary Level We would like to thank those who submitted the full papers and the reviewers for their timely help in assessing the papers for publication th We would also like to pay a special tribute to all the sponsors of the 20 ICCE and in particular the Tertiary Education Commission <http://tec.intnet.mu> and the Organisation for the Prohibition of Chemical Weapons <http://www.opcw.org> for kindly agreeing to fund the publication of these proceedings *Science Spectrum* Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2003-03

Computational Quantum Chemistry Charles M. Quinn, 2002-02-28 Computational Quantum Chemistry removes much of the mystery of modern computer programs for molecular orbital calculations by showing how to develop Excel spreadsheets to perform model calculations and investigate the properties of basis sets Using the book together with the CD ROM provides a unique interactive learning tool In addition because of the integration of theory with working examples on the CD ROM the reader can apply advanced features available in the spreadsheet to other applications in chemistry physics

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Hazard Analysis Techniques for System Safety Clifton A. Ericson, II,2015-07-20 Explains in detail how to perform the most commonly used hazard analysis techniques with numerous examples of practical applications Includes new chapters on Concepts of Hazard Recognition Environmental Hazard Analysis Process Hazard Analysis Test Hazard Analysis and Job Hazard Analysis Updated text covers introduction theory and detailed description of many different hazard analysis techniques and explains in detail how to perform them as well as when and why to use each technique Describes the components of a hazard and how to recognize them during an analysis Contains detailed examples that apply the

methodology to everyday problems **Class 10th Science Worksheet** , This book is as per the guidelines syllabus and marking scheme issued by CBSE for Class X The salient features of this workbook are The questions in the this book have been so designed that complete syllabus is covered This book help students to identify their weak areas and improve them Additional it will help students gain confidence The questions in the book are of varying difficulty level and will help students evaluate their reasoning analysis and understanding of the subject matter *Fundamentals of Physics* Chandan Sengupta, There are workbooks and study notes available in market in plenty Then also this workbook will provide more scope to students having aspirations to prosper Most of the questions incorporated in this workbook are from different levels of examinations duly conducted by different boards of studies This workbook will also provide an ample scope to students for accelerating their regularized studies Some of the worksheets are prepared along with supporting solution notes and related concet notes These questions are equally important for various examinations This workbook will provide additional support to fellow students of Standard 9 of National curriculum It has the core content from CBSE curriculum Additional resources from other streams of study are duly incorporated Fundamental Biomechanics of Sport and Exercise James

Watkins,2014-03-26 Fundamental Biomechanics of Sport and Exercise is an engaging and comprehensive introductory textbook that explains biomechanical concepts from first principles showing clearly how the science relates to real sport and exercise situations The book is divided into two parts The first provides a clear and detailed introduction to the structure and function of the human musculoskeletal system and its structural adaptations essential for a thorough understanding of human movement The second part focuses on the biomechanics of movement describing the forces that act on the human body and the effects of those forces on the movement of the body Every chapter includes numerous applied examples from sport and exercise helping the student to understand how mechanical concepts describe both simple and complex movements from running and jumping to pole vaulting or kicking a football In addition innovative worksheets for field and laboratory work are included that contain clear objectives a description of method data recording sheets plus a set of exemplary data and worked analysis Alongside these useful features are definitions of key terms plus review questions to aid student learning with detailed solutions provided for all numerical questions No other textbook offers such a clear easy to understand introduction to the fundamentals of biomechanics This is an essential textbook for any biomechanics course taken as part of degree programme in sport and exercise science kinesiology physical therapy sports coaching or athletic training

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