

Document Title: Description	Function Group: 000	Information Type: Service Information	Date: 27-Jan-21
Profile: L60E Volvo			

## Description

Showing Selected Profile

Valid for serial numbers:			
Model	Production site	Serial number start	Serial number stop
L60E Volvo			

Wheel loaders L60E, L70E och L90E have articulated frame steering and four-wheel drive.

The engine is a six-cylinder, four-stroke, direct-injection, turbocharged diesel engine of the type D6D.

The transmission is hydro-mechanical where all gears are in constant mesh. Its type designation is HT93 for L60E, HT95 for L70E and HTE202 for L90E.

There is a single-stage hydraulic torque converter between the engine and the transmission.

Front and rear axles have fully floating drive shafts with planetary gears in the wheel hubs. The front axle is equipped with a differential lock.

The service brake is a disc brake with disc brake pads in each wheel hub.

The parking brake on L60E and L70E is a disc brake positioned externally on the front output shaft of the transmission. On L90E the parking brake is positioned internally on the front output shaft of the transmission.

The steering system is hydrostatic with a variable load-sensing axial piston pump and two hydraulic cylinders (steering cylinders).

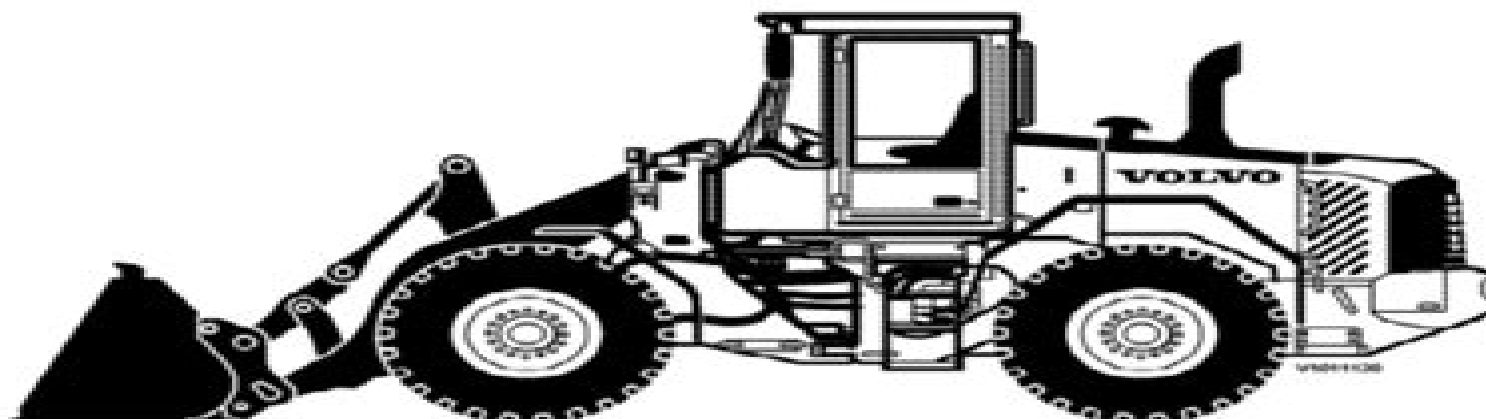


Figure 1

# Volvo L60e Loader Troubleshooting Guide

**Jex RM**



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web feb 24 2019 question 1 it is important to be able to recognize the major types of chemical reactions comstock getty images the chemical reaction  $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$  is a a synthesis reaction b decomposition reaction c single displacement reaction d double displacement reaction e combustion reaction question 2

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web an answer key document has been included and is available for download some additional information to be aware of for teachers demonstrations 1 and 5 will not produce a chemical reaction five of the reactions will produce indicators of a chemical reaction for students to observe

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web identifying chemical reaction types key part 1 identify the type of reaction 1  $\text{Na}_3\text{PO}_4 + 3\text{KOH} \rightarrow 3\text{NaOH} + \text{K}_3\text{PO}_4$  double displacement 2  $\text{MgCl}_2 + \text{Li}_2\text{CO}_3 \rightarrow \text{MgCO}_3 + 2\text{LiCl}$  double displacement 3  $\text{C}_6\text{H}_{12} + 9\text{O}_2 \rightarrow 6\text{CO}_2 + 6\text{H}_2\text{O}$  combustion 4  $\text{Pb} + \text{FeSO}_4 \rightarrow \text{PbSO}_4 + \text{Fe}$  single displacement 5  $\text{CaCO}_3 \rightarrow \text{CaO} + \text{CO}_2$  decomposition 6  $\text{P}_4 + 3\text{O}_2 \rightarrow 2\text{P}_2\text{O}_3$

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web chemical equations can also be used to represent physical processes write a chemical reaction for the boiling of water including the proper phase labels chemical equations can also be used to represent physical processes write a chemical reaction for the freezing of water including the proper phase labels explain why  $4\text{Na} + \text{S} \rightarrow 2\text{Na}_2\text{S}$

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web name score date balancing and identifying chemical reactions a identify each type of chemical reactions 1  $\text{CH}_4 + 2\text{O}_2 \rightarrow \text{CO}_2 + 2\text{H}_2\text{O}$  2  $\text{Sr} + 2\text{AgNO}_3 \rightarrow \text{Sr}(\text{NO}_3)_2 + 2\text{Ag}$  3  $2\text{LiBr} + \text{F}_2 \rightarrow 2\text{LiF} + \text{Br}_2$  4  $2\text{Na} + \text{F}_2 \rightarrow 2\text{NaF}$  5  $2\text{Br}_2 + \text{C} \rightarrow 2\text{CBr}_4$  b balance the chemical reactions and identify each type

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web 1  $3\text{NaBr} + \text{H}_3\text{PO}_4 \rightarrow \text{Na}_3\text{PO}_4 + 3\text{HBr}$  type of reaction double displacement 2  $3\text{Ca}(\text{OH})_2 + 2\text{Al}_2(\text{SO}_4)_3 \rightarrow 3\text{CaSO}_4 + 4\text{Al}(\text{OH})_3$  type of reaction double displacement 3  $3\text{Mg} + \text{Fe}_2\text{O}_3 \rightarrow 2\text{Fe} + 3\text{MgO}$  type of reaction single displacement 4  $\text{C}_2\text{H}_4 + 3\text{O}_2 \rightarrow 2\text{CO}_2 + 2\text{H}_2\text{O}$  type of reaction combustion 5  $2\text{PbSO}_4 + 2\text{PbSO}_3 + \text{O}_2$  type of reaction

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web a chemical reaction is a process in which one or more substances are converted to one or more different substances the starting substances are called the reactants and the new substances that form are called the products

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differentiate between a physical and chemical change physical does not change the identity of the matter can be changed  
back easily chemical new substance is created with new physical and chemical