Name: _____ Chamberlain
Class: _____

Lab: Types of Chemical Reactions

Purpose: A chemical reaction is a chemical change that takes place between two or more elements. There are five basic types of reactions that we will study in this lab: synthesis, decomposition, single replacement, double replacement, and combustion. This lab will introduce you to these reactions and how they work.

Synthesis Reactions

Station 1

Caution: This portion of the lab is very dangerous. Do not look directly at the burning metal. You must view it only through the blue or yellow glass. It will cause permanent blindness.

- Place a precut strip of magnesium in the flame over a bunsen burner.
- Record your observations below.
- Write the balanced chemical equation for this reaction below. (Hint: When you burned the magnesium, you were burning it in the air)
- 4. What evidence is there that a reaction took place?
- On Mrs. C's computer, Go to the website http://jchemed.chem.wisc.edu/jcesoft/cca/CCA0/Movies/NACL1.html. Watch the Quicktime video.
- 6. Record observations of the reaction in the space below:
- 7. Write the balanced equation for the reaction that takes place between sodium and chlorine.
- 8. Go to the website http://www.youtube.com/watch?v=N5h5ohd8298. Watch the video.
- Record observations of the reaction in the space below:
- 10. Write the balanced equation for the reaction between iodine and aluminum.

Types Of Chemical Reactions Lab 1answers

Brendan G. Carr

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