PERIODIC TABLE TRENDS

Step 1: Complete the squares for each element by adding the atomic number, name, and atomic mass. Write the atomic number at the top of the square, . Be Write the atomic name under the symbol. Write the atomic mass at the bottom of the square. Step 2: Determine the number of electrons, protons, Bohr Diagram Dot Structure and neutrons in each statement. Вe Step 3: Create a Bohr diagram for each element. -Metal Nonmetal Metalloid Step 4: Draw a Lewis Dot Structure for each element. Solid Cities.

Step 5: Show whether the element is a metal, nonmetal, or metalloid by circling the correct response. Also determine if the element is a solid or gas by circling the correct choice.

Step 6: Use the following colors to shade in the square for each element. You should ONLY color in the small square in the upper left-hand corner and not the entire card.

Green - Li and Na
Purple - F and Cl
Brown - N and P

Pink - O and S

Pink - O and S

Orange - B and Al
Yellow - He, Ne, and Ar

Blue - Be and Mg
Red - C and Si
White - H

Step 7: Cut the cards apart and arrange according to atomic number in the pattern shown below. Once you have the cards arranged in the correct order, glue them to a large sheet of construction paper.

 1
 2

 3
 4
 5
 6
 7
 8
 9
 10

 11
 12
 13
 14
 15
 16
 17
 18

Step 8: Put a title on your table.

Step 9: Make a key for each color with the name of the group for that column.

Step 10: Write the column number at the top of each group and the period number on the left side of the table. Be careful, column numbers are not in order from 1-18 and the transition metals, columns 3-12 are not on the table.

Step 11: Draw the red zigzag line that separates the metals from the nonmetals.

Step 12: Answer the questions on the worksheet using the information on your periodic table.

Periodic Table Assignment Answer Key

PT Brinkman

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College Chemistry I Norman H. Nachtrieb, 1960 Assessing Middle and High School Mathematics & Science Sheryn Spencer-Waterman, 2013-08-16 For middle and high school teachers of mathematics and science this book is filled with examples of instructional strategies that address students readiness levels interests and learning preferences It shows teachers how to formatively assess their students by addressing differentiated learning targets Included are detailed examples of differentiated formative assessment schedules plus tips on how to collaborate with others to improve assessment processes Teachers will learn how to adjust instruction for the whole class for small groups and for individuals They will also uncover step by step procedures for creating their own lessons infused with opportunities to formatively assess students who participate in differentiated learning activities Learning Chemistry 7 Solution Book (Year 2023-24), 2024-01-02 Science Scope, 2003 Learning Elementary Chemistry Class 7 Teacher Resource Book (Academic Year 2023-24) ,2023-05-20 Learning Elementary Chemistry Class 7 Teacher Resource Book Academic Year 2023 24 THE Journal ,1999 College Chemistry II Norman H. Nachtrieb. 1960 **Holt Chemistry** Ralph Thomas Myers, 2004 **Inorganic** Chemistry-II (For M.Sc. Course for Universities in Uttarakhand) Dr. Geeta Tiwari, This book entitled Inorganic Chemistry II is an effort to present the subject matter in a comprehensible and easily understandable form This textbook is purposefully prepared for the postgraduate Inorganic Chemistry second semester course and it covers all the topics recommended Eye Tracking for STEM Education Research: New Perspectives Pascal Klein, Martin Rusek, Maike Schindler, 2024-04-04 A modern approach to improving education uses the components of experimental scientific research practices based on objective data dissemination of results and the use of modern technologies STEM education research is maturing and new tools and analysis techniques become available As one example eye tracking the recording of persons eye movements has been growing in popularity as it enables researchers to study learning materials effectiveness problem solving and even students approaches during experimentation Eye movements as captured using eye tracking can reveal information about a student s attention and cognition on a process level going well beyond classical product based assessment techniques such as questionnaires or tests Study Guide Woolfolk, 2000-06 **Biology/science Materials** Armchair Chemistry David Bradley, Joel Levy, 2018-02-13 Part of the Armchair Carolina Biological Supply Company, 1991 series Armchair Chemistry is a quick refresher course in how we survey of the science It explains how we evolved from believing in alchemy to discovering modern chemical equations and goes into detail about the lives of the scientists that uncovered them Fascinating and interactive this is ideal for the student brushing up on a subject or for as a clear and accessible companion for beginner s and experts alike It contains explanations of different chemical concepts as well as profiles of key scientists and and their discoveries It contains clear and concise explanations of different chemical concepts as well as profiles of key scientists and their discoveries A unique feature of the book is its simple step by step exercises

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