

Periodic Table Extra Practice Worksheet

Directions: Answer the following questions using the Periodic Table below.

IA 1 H 1							VIII 2 He 4
	IIA	IIIB	IVB	VB	VIB	VIIIB	
3 Li 7	4 Be 9	5 B 11	6 C 12	7 N 14	8 O 16	9 F 19	10 Ne 20
11 Na 23	12 Mg 24	13 Al 27	14 Si 28	15 P 31	16 S 32	17 Cl 35	18 Ar 40

1. How many protons, neutrons, and electrons are in each of the following atoms?

Element	Boron (B)	Sodium (Na)	Phosphorus (P)
Protons	5	11	15
Neutrons	6	12	16
Electrons	5	11	15

2. Which diagram best shows nucleus of beryllium? Circle the correct answer.

n = 4
e = 5

A

p = 4
e = 4

B

p = 5
n = 4

C

p = 4
n = 5

D

n = 5
e = 4

E

p = 9
n = 4

F

THE ANSWER IS D

3. Fill in the chart below from the data provided.

ELEMENT	ATOMIC NUMBER	MASS NUMBER	NUMBER OF PROTONS	NUMBER OF NEUTRONS	NUMBER OF ELECTRONS
Aluminum	a.13	27	b.13	c.14	d.13
Hydrogen	e.1	f.1	1	g.0	h.1
Argon	i.18	j.40	k.18	l.22	18
Magnesium	12	m.24	n.12	o.12	p.12

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A red circular graphic with a gradient, appearing as a partial circle or a thick curved line, located to the right of the 'Disha Experts' text.

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