

Stoichiometry Worksheet 2: Percent Yield

For each of the problems below:

- Write the balanced chemical equation
- Identify the given (with units) and what you want to find (with units)
- Show set up with units. Check sig figs, give final answer with units and label.

1. Using the Hoffman apparatus for electrolysis, a chemist decomposes 36 g of water into its gaseous elements. How many grams of hydrogen gas should she get (theoretical yield)?



Before 2 mol 0 mol 0 mol

Change -2 mol +2 mol 0 mol

After 0 mol 2 mol 0 mol

$$36 \text{ g H}_2\text{O} \times \frac{1 \text{ mol H}_2\text{O}}{18.0 \text{ g H}_2\text{O}} = 2.0 \text{ mol H}_2\text{O} \times \frac{2 \text{ mol H}_2}{2 \text{ mol H}_2\text{O}} = 2 \text{ mol H}_2 \times \frac{2.0 \text{ g H}_2}{1 \text{ mol H}_2} = 4.0 \text{ g H}_2$$

2. Recall that liquid sodium reacts with chlorine gas to produce sodium chloride. You want to produce 581 g of sodium chloride. How many grams of sodium are needed?



Before 9.94 mol xs mol 0 mol

Change -9.94 mol 0 mol +9.94 mol

After 0 mol 0 mol 9.94 mol

$$581 \text{ g NaCl} \times \frac{1 \text{ mol NaCl}}{58.5 \text{ g NaCl}} = 9.93 \text{ mol NaCl} \times \frac{2 \text{ mol Na}}{2 \text{ mol NaCl}} = 9.93 \text{ mol Na} \times \frac{23.0 \text{ g Na}}{1 \text{ mol Na}} = 228 \text{ g Na}$$

3. You eat 180.0 g of glucose (90 M&Ms). If glucose, $\text{C}_6\text{H}_{12}\text{O}_6$, reacts with oxygen gas to produce carbon dioxide and water, how many grams of oxygen will you have to breathe in to burn the glucose?



Before 1.0 mol xs mol 0 mol 0 mol

Change -1.0 mol -6.0 mol +6.0 mol +6.0 mol

After 0 mol xs mol 6.0 mol 6.0 mol

$$180.0 \text{ g C}_6\text{H}_{12}\text{O}_6 \times \frac{1 \text{ mol C}_6\text{H}_{12}\text{O}_6}{180 \text{ g C}_6\text{H}_{12}\text{O}_6} = 1.00 \text{ mol C}_6\text{H}_{12}\text{O}_6 \times \frac{6 \text{ mol O}_2}{1 \text{ mol C}_6\text{H}_{12}\text{O}_6} = 6.00 \text{ mol O}_2$$

$$6.0 \text{ mol O}_2 \times \frac{32.0 \text{ g O}_2}{1 \text{ mol O}_2} = 192.0 \text{ g O}_2$$

Percent Yield Stoichiometry Answers With Work

RD Boyd



Percent Yield Stoichiometry Answers With Work:

Whispering the Techniques of Language: An Emotional Quest through **Percent Yield Stoichiometry Answers With Work**

In a digitally-driven world wherever displays reign supreme and quick communication drowns out the subtleties of language, the profound techniques and emotional nuances hidden within phrases often move unheard. However, set within the pages of **Percent Yield Stoichiometry Answers With Work** a fascinating literary prize blinking with natural feelings, lies an extraordinary quest waiting to be undertaken. Penned by a skilled wordsmith, that marvelous opus invites viewers on an introspective trip, delicately unraveling the veiled truths and profound influence resonating within ab muscles fabric of every word. Within the emotional depths of this poignant evaluation, we will embark upon a sincere exploration of the book is key subjects, dissect their charming publishing design, and succumb to the strong resonance it evokes heavy within the recesses of readers hearts.

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