

# GRAPHING TRENDS OF THE PERIODIC TABLE

## CHEMISTRY

NAME: \_\_\_\_\_

DATE: \_\_\_\_\_

PERIOD: \_\_\_\_\_

### Graphing Periodic Table Trends

**Directions:** Using the provided graph paper, graph each of the elemental properties using a line graph. Place the property on the Y-axis and the Atomic number on the X-axis. Draw a dark, solid, vertical line on each graph where each row on the periodic table ends, (i.e. after atomic number 10). Use the graphs to analyze the trends in these properties and to answer the questions on the back.

Atomic Radius (pm)							
Lithium - 3	152	Sodium - 11	190	Potassium - 19	243	Rubidium - 37	265
Beryllium - 4	112	Magnesium - 12	145	Calcium - 20	194	Strontium - 38	219
Boron - 5	87	Aluminum - 13	118	Gallium - 31	136	Indium - 49	156
Carbon - 6	67	Silicon - 14	111	Germanium - 32	125	Tin - 50	145
Nitrogen - 7	56	Phosphorus - 15	98	Arsenic - 33	114	Antimony - 51	133
Oxygen - 8	48	Sulfur - 16	88	Selenium - 34	103	Tellurium - 52	123
Fluorine - 9	42	Chlorine - 17	79	Bromine - 35	94	Iodine - 53	115

Ionization Energy (kJ/mol)							
Lithium - 3	520	Sodium - 11	496	Potassium - 19	419	Rubidium - 37	403
Beryllium - 4	900	Magnesium - 12	738	Calcium - 20	590	Strontium - 38	550

### Atomic Radius



# Trends In The Periodic Table Graphing

**Yan Bai**



## Trends In The Periodic Table Graphing:

**Computer Based Projects for a Chemistry Curriculum** Thomas J. Manning, Aurora P. Gramatges, 2013-04-04 This e book is a collection of exercises designed for students studying chemistry courses at a high school or undergraduate level. The e book contains 24 chapters each containing various activities employing applications such as MS excel spreadsheets and Spartan computational modeling. Each project is explained in a simple easy to understand manner. The content within this book is suitable as a guide for both teachers and students and each chapter is supplemented with practice guidelines and exercises. Computer Based Projects for a Chemistry Curriculum therefore serves to bring computer based learning a much needed addition in line with modern educational trends to the chemistry classroom. Basic Concepts of Chemistry Leo J. Malone, Theodore O. Dolter, 2011-12-27 The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment. New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections, assessment exercises at the end of each section and relevant chapter problems at the end of each chapter. Every concept in the text is clearly illustrated with one or more step by step examples. Making it Real essays have been updated to present timely and engaging real world applications emphasizing the relevance of the material they are learning. This edition continues the end of chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter. WileyPLUS sold separately from text. Science Units for Grades 9-12 Randy L. Bell, Joe Garofalo, 2005 Sample topics include cell division, virtual dissection, earthquake modeling, the Doppler Effect and more. *Chemistry Homework for OCR A for Double and Separate Awards* Gareth Pritchard, 2001 Homework activities for OCR A Chemistry specifications. The Science Teachers Bulletin, 1989 *Handbook of Research on K-12 Blended and Virtual Learning Through the i<sup>2</sup>Flex Classroom Model* Avgerinou, Maria D., Pelonis, Peggy, 2021-03-05 Teaching models that focus on blended and virtual learning have become important during the past year and have become integral for the continuance of learning. The i Flex classroom model, a variation of blended learning, allows non-interactive teaching activities to take place without teachers' direct involvement, freeing up time for more meaningful teacher-student and student-student interactions. There is evidence that i Flex leads to increased student engagement and motivation as well as better exploitation of teachers and classroom time, leading to the development of higher order cognitive skills as well as study skills for students' future needs related to citizenship, college and careers. The Handbook of Research on K-12 Blended and Virtual Learning Through the i Flex Classroom Model focuses not only on how to design, deliver and evaluate courses but also on how to assess teacher performance in a blended i<sup>2</sup>Flex way at the K-12 level. The book will discuss the implementation of the i Flex iSquareFlex, a non-traditional learning methodology which integrates internet-based delivery of content and instruction with faculty-guided student-independent learning in combination with face-to-face classroom instruction, aiming at

developing higher order cognitive skills within a flexible learning design framework While highlighting new methods for improving the classroom and learning experience in addition to preparing students for higher education and careers this publication is an essential reference source for pre service and in service teachers researchers administrators educational technology developers and students interested in how the i2Flex model was implemented in classrooms and the effects of this learning model

**An Introduction to Chemistry** Michael Mosher,Paul Kelter,2023-03-18 This textbook is written to thoroughly cover the topic of introductory chemistry in detail with specific references to examples of topics in common or everyday life It provides a major overview of topics typically found in first year chemistry courses in the USA The textbook is written in a conversational question based format with a well defined problem solving strategy and presented in a way to encourage readers to think like a chemist and to think outside of the box Numerous examples are presented in every chapter to aid students and provide helpful self learning tools The topics are arranged throughout the textbook in a traditional approach to the subject with the primary audience being undergraduate students and advanced high school students of chemistry

**Exploring Chemical Concepts Through Theory and Computation** Shubin Liu,2024-05-15 Exploring Chemical Concepts Through Theory and Computation Deep theoretical resource on the essence of chemistry explaining a variety of important concepts including redox states and bond types Exploring Chemical Concepts Through Theory and Computation provides a comprehensive account of how the three widely used theoretical frameworks of valence bond theory molecular orbital theory and density functional theory along with a variety of important chemical concepts can between them describe and efficiently and reliably predict key chemical parameters and phenomena By comparing the three main theoretical frameworks readers will become competent in choosing the right modeling approach for their task The authors go beyond a simple comparison of existing algorithms to show how data driven theories can explain why chemical compounds behave the way they do thus promoting a deeper understanding of the essence of chemistry The text is contributed to by top theoretical and computational chemists who have turned computational chemistry into today s data driven and application oriented science Exploring Chemical Concepts Through Theory and Computation discusses topics including Orbital based approaches density based approaches chemical bonding partial charges atoms in molecules oxidation states aromaticity and antiaromaticity and acidity and basicity Electronegativity hardness softness HSAB sigma hole interactions charge transport and energy transfer and homogeneous and heterogeneous catalysis Electrophilicity nucleophilicity cooperativity frustration homochirality and energy decomposition Chemical concepts in solids excited states spectroscopy and machine learning and catalysis and machine learning as well as key connections between related concepts Aimed at both novice and experienced computational theoretical and physical chemists Exploring Chemical Concepts Through Theory and Computation is an essential reference to gain a deeper more advanced holistic understanding of the field of chemistry as a whole

*Jacaranda Chemistry 1 VCE Units 1 and 2, learnON and Print* Neale Taylor,Angela Stubbs,Robert Stokes,2022-11-30 Developed by

expert Victorian teachers for VCE students The NEW Jacaranda Chemistry VCE series continues to deliver curriculum aligned material that caters to students of all abilities Our expert author team of practising teachers and assessors ensures 100% coverage of the new VCE Chemistry Study Design 2023 2027 **Chemistry 'O' Level Guide , Jacaranda Core Science Stage 4 New South Wales Australian Curriculum, 3e learnON and Print** Paul Arena,2025-08-25

Intermetallic Chemistry Riccardo Ferro,Adriana Saccone,2011-08-26 Intermetallic science is closely related to physics chemistry metallurgy materials science technology and engineering This book emphasizes the chemical aspects of this science and therefore the mutual reactivity of metals and the characteristics of intermetallic compounds Topics included are Phase diagrams of alloy systems Many intermetallic systems form several compounds generally not obeying common simple stoichiometric rules which are often homogeneous in a certain range of compositions The stability and extension of these phases are conveniently presented through phase diagrams Selected aspects of intermetallics structural chemistry with emphasis on the solid state The general structural characteristics of intermetallic phases are considered with attention to nomenclature and to alternative and complementary methods of presenting crystal chemical data A brief account is given of derivative and degenerate structures modular aspects of crystal structures and of a few special groups of alloys such as quasicrystals and amorphous alloys A number of selected structural prototypes with typical features their possible grouping in structural families and their distribution among different types of alloys are provided Intermetallic reactivity trends in the Periodic Table Attention is given to a few selected elemental parameters such as electron configuration and valence electron number and to their changes along the Table which act as reference factors of the intermetallic behaviour As an example the relationships are considered between crystal structure and the number of valence electrons per atom or per formula in various classes of compounds or solid solution phases Alloying behaviour systematics of intermetallic systems with a description of the intermetallic reactivity of each element or group of elements in the order of their position in the Periodic Table For each pair of metallic elements their capability to form intermediate phases is summarised by maps and schemes A description of small scale preparation methods of intermetallics A number of interesting and significant peculiarities are e g those related to their high melting points insolubility in common solvents etc Systematic treatment of alloying behaviour Wide overview of intermetallic chemistry Illustrated with many examples AP Chemistry Premium, 2022-2023: Comprehensive Review with 6 Practice Tests + an Online Timed Test Option Neil D. Jespersen,Pamela Kerrigan,2021-07-06 Be prepared for exam day with Barron s Trusted content from AP experts Barron s AP Chemistry Premium 2022 2023 includes in depth content review and online practice It s the only book you ll need to be prepared for exam day Written by Experienced Educators Learn from Barron s all content is written and reviewed by AP experts Build your understanding with comprehensive review tailored to the most recent exam Get a leg up with tips strategies and study advice for exam day it s like having a trusted tutor by your side Be Confident on Exam Day Sharpen your test taking skills with 6 full length practice

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*Ebook: Chemistry* Julia Burdge,2014-10-16 Chemistry Third Edition by Julia Burdge offers a clear writing style written with the students in mind Julia uses her background of teaching hundreds of general chemistry students per year and creates content to offer more detailed explanation on areas where she knows they have problems With outstanding art a consistent problem solving approach interesting applications woven throughout the chapters and a wide range of end of chapter problems this is a great third edition text

**Springer Handbook of Glass** J. David Musgraves,Juejun Hu,Laurent Calvez,2019-11-08 This handbook provides comprehensive treatment of the current state of glass science from the leading experts in the field Opening with an enlightening contribution on the history of glass the volume is then divided into eight parts The first part covers fundamental properties from the current understanding of the thermodynamics of the amorphous state kinetics and linear and nonlinear optical properties through colors photosensitivity and chemical durability The second part provides dedicated chapters on each individual glass type covering traditional systems like silicates and other oxide systems as well as novel hybrid amorphous materials and spin glasses The third part features detailed descriptions of modern characterization techniques for understanding this complex state of matter The fourth part covers modeling from first principles calculations through molecular dynamics simulations and statistical modeling The fifth part presents a range of laboratory and industrial glass processing methods The remaining parts cover a wide and representative range of applications areas from optics and photonics through environment energy architecture and sensing Written by the leading international experts in the field the Springer Handbook of Glass represents an invaluable resource for graduate students through academic and industry researchers working in photonics optoelectronics materials science energy architecture and more

**The Johnstone Triangle** Norman Reid,2021-01-29 Chemistry is often seen as a difficult subject to understand This book focusses on the triangle model that Alex H Johnstone developed in the early 1980s The model has been applied in almost every area of education in chemistry at all stages of learning

**Chemistry for AQA.** Ann Fullick,Patrick Fullick,2001 This resource has separate books for biology chemistry and physics Each book is accompanied by a teacher s resource pack on customizable CD ROM or as a printed pack The series is designed to work in conjunction with the Coordinated Science for AQA series so that coordinated and separate science can be taught alongside each other

**Chemistry for the IB MYP 4 & 5** Annie Termaat,Christopher Talbot,2016-08-22 The only series for MYP 4 and 5 developed in cooperation with the International Baccalaureate IB Develop your skills to become an inquiring learner ensure you navigate the MYP framework with confidence using a concept driven and assessment focused

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