FINAL EXAM Summary Sheet

Potential Study Questions

<u>Disclaimer</u>: The following are potential topics for exam questions and sample questions; the exam may or may not contain these questions, and questions not included in this sheet may or may not be on the exam. There also may be typos in this sheet.

ALKANES

Physical Properties

- Hydrocarbons, dispersion forces, non-polar, not miscible w/water, low density
- sp³ hybridized carbon: C-C, C-H bonds, electronegativity, MO theory & orbitals
- m.p.: odd/even effect.
- b.p. C1-4: gases, C5-17: liquids, C18 and up: solids
- linear, branched, cyclics, chirality (incl. optical activity, D/L, R/S, d/l,+/-),
- non-equivalent carbons
- isomers: constitutional, conformational isomers, stereo-isomers
- naming, special groups:
 - o Me, Et, Pr, Bu
 - o iPr, iBu, sec-Bu, tBu
 - o neopentyl

<u>Task</u>: Name compounds 1 and 2; identify any chiral centers and # of chemically non-equivalent carbons; where would the compounds absorb in IR in UVvis? Identify the participating orbitals.



- Rotational/conformational isomers = conformations
 - o eclipsed
 - values: H/H, CH3/H, CH3/CH3
 - o staggered, anti, gauche
 - value: gauche: CH3/CH3

Task: draw an energy rotational energy plot (360°) for pentane along C2-C3 bond

- Cycloalkanes:
 - o Spiro, fused, bridged: naming
 - o Heats of combustion: ring strains (angle, torsional, sterics)
 - o Cis/trans
 - 1,3-diaxial interactions
 - value H/CH3
 - Ring-flip for cyclohexane

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JE Gale

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Table of Contents Organic Chemistry Final Exam Study Guide

- 1. Understanding the eBook Organic Chemistry Final Exam Study Guide
 - The Rise of Digital Reading Organic Chemistry Final Exam Study Guide
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Organic Chemistry Final Exam Study Guide
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Organic Chemistry Final Exam Study Guide
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Organic Chemistry Final Exam Study Guide
 - Personalized Recommendations
 - o Organic Chemistry Final Exam Study Guide User Reviews and Ratings
 - Organic Chemistry Final Exam Study Guide and Bestseller Lists
- 5. Accessing Organic Chemistry Final Exam Study Guide Free and Paid eBooks

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- o Organic Chemistry Final Exam Study Guide Budget-Friendly Options
- 6. Navigating Organic Chemistry Final Exam Study Guide eBook Formats
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- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Organic Chemistry Final Exam Study Guide
 - Highlighting and Note-Taking Organic Chemistry Final Exam Study Guide
 - Interactive Elements Organic Chemistry Final Exam Study Guide
- 8. Staying Engaged with Organic Chemistry Final Exam Study Guide
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Organic Chemistry Final Exam Study Guide
- 9. Balancing eBooks and Physical Books Organic Chemistry Final Exam Study Guide
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Organic Chemistry Final Exam Study Guide
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Organic Chemistry Final Exam Study Guide
 - Setting Reading Goals Organic Chemistry Final Exam Study Guide
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Organic Chemistry Final Exam Study Guide
 - Fact-Checking eBook Content of Organic Chemistry Final Exam Study Guide
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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